

Stoichiometry of Na⁺-HCO₃⁻ cotransport in basolateral membrane vesicles isolated from rabbit renal cortex.

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Research Article

The major pathway for HCO₃⁻ transport across the basolateral membrane of the proximal tubule cell is electrogenic Na⁺-HCO₃⁻ cotransport. In this study, we have determined the stoichiometry of the Na⁺-HCO₃⁻ cotransport system in basolateral membrane vesicles that were isolated from rabbit renal cortex by Percoll gradient centrifugation. When the membrane potential is approximated by the Nernst potential for K⁺, as in the presence of the K⁺ ionophore valinomycin, equilibrium thermodynamics predicts that the Na⁺-HCO₃⁻ cotransport system should come to equilibrium and mediate no net flux when $(\text{Na})_i/(\text{Na})_o = [(\text{HCO}_3)_o/(\text{HCO}_3)_i]^n[(\text{K})_o/(\text{K})_i]^{n-1}$, where n is the HCO₃⁻:Na⁺ stoichiometry. Our experimental approach was to impose transmembrane Na⁺, HCO₃⁻, and K⁺ gradients of varying magnitude and direction, and then to measure the net flux of Na⁺ over the subsequent 3-s period. In this way, we could determine the conditions for equilibrium of the transport system and thereby calculate n . The results of these experiments indicate that the value of n is greater than 2.6 and less than 3.5, consistent with a stoichiometry of 3 HCO₃⁻:1 Na⁺, or a thermodynamically equivalent process. Based on reported intracellular potentials and ion activities, this value for the stoichiometry indicates that the inside-negative membrane potential is sufficient to drive HCO₃⁻ exit against the inward concentration gradients of HCO₃⁻ and Na⁺ that are present across the basolateral membrane of the intact [...]

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Stoichiometry of $\text{Na}^+ - \text{HCO}_3^-$ Cotransport in Basolateral Membrane Vesicles Isolated from Rabbit Renal Cortex

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Abstract

The major pathway for HCO_3^- transport across the basolateral membrane of the proximal tubule cell is electrogenic $\text{Na}^+ - \text{HCO}_3^-$ cotransport. In this study, we have determined the stoichiometry of the $\text{Na}^+ - \text{HCO}_3^-$ cotransport system in basolateral membrane vesicles that were isolated from rabbit renal cortex by Percoll gradient centrifugation. When the membrane potential is approximated by the Nernst potential for K^+ , as in the presence of the K^+ ionophore valinomycin, equilibrium thermodynamics predicts that the $\text{Na}^+ - \text{HCO}_3^-$ cotransport system should come to equilibrium and mediate no net flux when $(\text{Na})_i/(\text{Na})_o = [(\text{HCO}_3)_o/(\text{HCO}_3)_i]^n[(\text{K})_o/(\text{K})_i]^{n-1}$, where n is the $\text{HCO}_3^-:\text{Na}^+$ stoichiometry. Our experimental approach was to impose transmembrane Na^+ , HCO_3^- , and K^+ gradients of varying magnitude and direction, and then to measure the net flux of Na^+ over the subsequent 3-s period. In this way, we could determine the conditions for equilibrium of the transport system and thereby calculate n . The results of these experiments indicate that the value of n is > 2.6 and < 3.5 , consistent with a stoichiometry of 3 $\text{HCO}_3^-:1 \text{ Na}^+$, or a thermodynamically equivalent process. Based on reported intracellular potentials and ion activities, this value for the stoichiometry indicates that the inside-negative membrane potential is sufficient to drive HCO_3^- exit against the inward concentration gradients of HCO_3^- and Na^+ that are present across the basolateral membrane of the intact proximal tubule cell under physiologic conditions.

Introduction

~ 90% of the filtered load of HCO_3^- is reabsorbed in the proximal tubule of the kidney by a process of active acid secretion. The principal pathway for uphill movement of acid from cell to tubular fluid is via the luminal membrane $\text{Na}^+ - \text{H}^+$ exchanger (1). The pathway(s) for exit of HCO_3^- across the basolateral membrane of the proximal tubule cell have been less well characterized. Boron and Boulpaep described a process of $\text{Na}^+ - \text{HCO}_3^-$ cotransport across the basolateral membrane of the amphibian

proximal tubule (2). Recent studies suggest that this transport process is also the major mechanism for exit of HCO_3^- across the basolateral membrane of the mammalian proximal tubule cell (3–10). In addition, $\text{Na}^+ - \text{HCO}_3^-$ cotransport has been observed in corneal endothelial cells (11, 12), suggesting that this pathway is involved in mediating HCO_3^- transport in epithelial tissues other than the kidney.

In all cases studied to date, $\text{Na}^+ - \text{HCO}_3^-$ cotransport has been found to be electrogenic and associated with a net flux of negative charge (2, 3, 5–8, 11). This property indicates that the stoichiometry of the cotransport process must involve more than 1 $\text{HCO}_3^-:\text{Na}^+$. Knowledge of the precise stoichiometry is important for predicting the direction of net transport under physiologic conditions. The greater the $\text{HCO}_3^-:\text{Na}^+$ stoichiometry, and hence the greater the net negative charge movement per transport event, the more effectively the inside-negative membrane potential of the cell can act as a driving force to bring about the net exit of HCO_3^- against the Na^+ gradient normally directed inward across the plasma membrane. For example, recent measurements (13) indicate that the membrane potential would not be sufficient to drive net HCO_3^- efflux across the basolateral membrane of the rat proximal tubule cell under physiologic conditions if the stoichiometry of cotransport were only 2 $\text{HCO}_3^-:\text{Na}^+$ or 1 $\text{CO}_3^{2-}:\text{Na}^+$, as has been generally assumed (2, 10, 11).

Recently, Yoshitomi et al. (5) estimated the stoichiometry of $\text{Na}^+ - \text{HCO}_3^-$ cotransport across the basolateral membrane of the intact rat proximal tubule cell. These investigators calculated a value of 3, based on comparison of the fall in both intracellular Na^+ activity and intracellular pH that occur in response to sudden reduction of peritubular HCO_3^- concentration. However, the reliability of this value is uncertain because other basolateral membrane transport systems involved in basolateral membrane Na^+ transport, such as the Na,K-ATPase , are known to be markedly pH-dependent (14). Moreover, calculation of the HCO_3^- flux corresponding to a measured change in intracellular pH depends critically on estimation of intracellular buffering capacity, which is difficult to determine precisely. Indeed, the same investigators have reported a two-fold range of values for the intracellular buffering capacity in rat proximal tubule cells (13, 15).

Accordingly, we have attempted to determine the stoichiometry of the $\text{Na}^+ - \text{HCO}_3^-$ cotransport system by a method not subject to these uncertainties. We have used a thermodynamic approach to study this problem in basolateral membrane vesicles isolated from the rabbit renal cortex. We previously demonstrated that electrogenic $\text{Na}^+ - \text{HCO}_3^-$ cotransport is a major pathway for transport of Na^+ and HCO_3^- in these membrane vesicles (16). We find that the stoichiometry of $\text{Na}^+ - \text{HCO}_3^-$ cotransport is 3 $\text{HCO}_3^-:\text{Na}^+$ or a thermodynamically equivalent process.

A preliminary report of this work was previously published as an abstract (1986. *Clin. Res.* 34:609A).

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Methods

Basolateral membrane vesicles were isolated from rabbit renal cortex by differential and Percoll gradient centrifugation, as recently described (16). The membrane vesicles were suspended in a medium consisting of 250 mM sucrose, 10 mM Hepes titrated to pH 7.6 with tetramethylammonium (TMA)⁺ hydroxide, and then frozen and stored at -70°C until used.

Intravesicular content of ^{22}Na (0.16–0.25 mg membrane protein per sample) was assayed in triplicate by rapid filtration, as previously described (16). The ice-cold medium used to dilute and wash the vesicles consisted of 170 mM K gluconate, 10 mM Hepes titrated to pH 7.5 with TMA hydroxide. In general, net fluxes of ^{22}Na were assayed over a 3-s interval. Na^+ uptake with time in the presence of an imposed HCO_3^- gradient is linear with time for over 3 s in rabbit renal basolateral membrane vesicles (16), indicating that dissipation of imposed ion gradients is minimal within this time interval. Other details of the experimental protocols are described in the figure legends.

We purchased ^{22}Na from Amersham Corp., Arlington Heights, IL, valinomycin from Sigma Chemical Co., St. Louis, MO, and Percoll from Pharmacia Inc., Piscataway, NJ. Valinomycin was added to the membrane suspension in a 1:100 dilution from a stock solution in 95% ethanol.

Results and Discussion

The stoichiometry of a coupled transport process can be determined by finding the electrochemical potential difference of one solute that just balances the electrochemical potential difference of a second solute, thereby bringing the transport process to equilibrium (17). This approach has previously been used to determine the stoichiometry of $\text{Na}^+ - \text{H}^+$ exchange (18) and Na^+ -glucose cotransport (19) in luminal membrane vesicles isolated from the rabbit renal cortex. For the case of $\text{Na}^+ - \text{HCO}_3^-$ cotransport, the transport system will be at equilibrium and mediate no net flux when:

$$\Delta\tilde{\mu}_{\text{Na}}^{i \rightarrow o} = n\Delta\tilde{\mu}_{\text{HCO}_3}^{o \rightarrow i} \quad (1)$$

where $\Delta\tilde{\mu}_{\text{Na}}^{i \rightarrow o}$ is the in-to-out electrochemical potential difference for Na^+ , $\Delta\tilde{\mu}_{\text{HCO}_3}^{o \rightarrow i}$ is the out-to-in electrochemical potential difference for HCO_3^- , and n is the number of HCO_3^- cotransported per Na^+ (17). Expressing the electrochemical potential differences in terms of the relevant inside and outside ion concentrations and the membrane potential, V_m , yields:

$$\begin{aligned} &[(\text{Na})_i/(\text{Na})_o] \exp(FV_m/RT) \\ &= \{[(\text{HCO}_3)_o/(\text{HCO}_3)_i] \exp(FV_m/RT)\}^n. \end{aligned} \quad (2)$$

In the presence of the K^+ ionophore valinomycin, the membrane potential V_m will approach the Nernst potential for K^+ :

$$V_m = (RT/F) \ln [(K)_o/(K)_i]. \quad (3)$$

Combining and rearranging Eq. 2 and 3 gives:

$$(\text{Na})_i/(\text{Na})_o = [(\text{HCO}_3)_o/(\text{HCO}_3)_i]^n [(K)_o/(K)_i]^{n-1}. \quad (4)$$

Based on this last equation, the general experimental strategy for calculating n was to set $(K^+)_o = (K^+)_i$ and find the conditions under which an inward HCO_3^- gradient would just balance an outward Na^+ gradient so as to bring net flux through the system to zero, or to set $(\text{HCO}_3^-)_o = (\text{HCO}_3^-)_i$ and find the conditions under which an inward K^+ gradient would just balance an outward Na^+ gradient so as to bring net flux to zero.

The results of an experiment utilizing the first approach are illustrated in Fig. 1. Membrane vesicles were preloaded with ^{22}Na and then diluted 10-fold so that a 10:1 outward Na^+ gradient was imposed. The composition of the diluting media was adjusted to impose different inward HCO_3^- gradients while maintaining $(K^+)_o = (K^+)_i$ in the presence of valinomycin. The net change in intravesicular content of ^{22}Na was then assayed over the 3-s period immediately following the imposition of these ion gradients. Over this time interval, no net flux of ^{22}Na could be detected in the absence of a $\text{CO}_2/\text{HCO}_3^-$ buffer system. Thus, under the conditions of this and the following experiments, the measured net fluxes of Na^+ entirely represent HCO_3^- -dependent transport (i.e., $\text{Na}^+ - \text{HCO}_3^-$ cotransport). As shown in Fig. 1, imposing a 3.2:1 inward HCO_3^- gradient induced the net influx of Na^+ against its own 10:1 outward gradient. According to Eq. 4, if $n = 2$, the cotransport system would have been at equilibrium and no net flux would have been observed under these conditions. The fact that the 3.2:1 inward HCO_3^- gradient could overcome the 10:1 outward Na^+ gradient as a driving force indicates that n must be > 2 . When a 2.2:1 inward HCO_3^- gradient was imposed, no net Na^+ flux was observed, consistent with $n = 3$. Finally, when an inward HCO_3^- gradient of only 1.2:1 was imposed, net Na^+ efflux down its concentration gradient was observed. The value of n would have had to be 12 for this to be the condition for equilibrium. Thus, the results of this experiment indicate that the $\text{HCO}_3^- : \text{Na}^+$ stoichiometry must be $> 2:1$ and

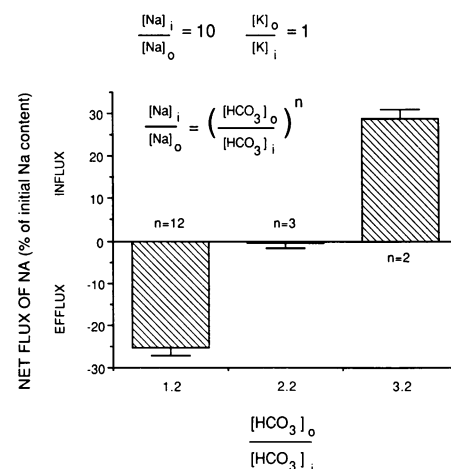


Figure 1. Net Na^+ flux in the presence of a 10:1 outward Na^+ gradient and varying inward HCO_3^- gradients. Basolateral membrane vesicles were pre-equilibrated for 120 min at 20°C in a medium that consisted of 1 mM ^{22}Na -gluconate, 52 mM TMA-gluconate, 42 mM Hepes, 23 mM TMA-OH, gassed with 90% N_2 , 10% CO_2 , and that contained in addition either 38 mM mannitol, 14 mM Mes, 22 mM K-HCO₃, and 55 mM K-gluconate, pH 7.08; 43 mM mannitol, 9 mM Mes, 32 mM K-HCO₃, and 45 mM K-gluconate, pH 7.24; or 50 mM mannitol, 2 mM Mes, 59 mM K-HCO₃, and 18 mM K-gluconate, pH 7.51; corresponding to inward HCO_3^- gradients of 3.2, 2.2 or 1.2:1, respectively. Intravesicular content of ^{22}Na was then measured before and 3 s after 1:10 dilution and incubation of the vesicles in a medium of final composition 53 mM TMA-gluconate, 42 mM Hepes, 51 mM mannitol, 1 mM Mes, 24 mM TMA-OH, 71 mM K-HCO₃, and 6 mM K-gluconate, pH 7.60, gassed with 90% N_2 , 10% CO_2 . Values shown for net flux represent means \pm SE for three separate experiments performed in triplicate on three different membrane preparations. For each imposed HCO_3^- gradient, the value of n is shown that would satisfy the illustrated form of Eq. 4, which applies when $(K^+)_o = (K^+)_i$.

1. Abbreviations used in this paper: TMA, tetramethylammonium.

< 12:1, with equilibrium observed under conditions corresponding to $n = 3$.

The goal of the next experiment (Fig. 2) was to define the stoichiometry of cotransport more precisely. In this case, vesicles were preloaded with ^{22}Na and then different outward Na^+ gradients were imposed by diluting 1 vol of Na^+ -loaded vesicles in different volumes of Na^+ -free media. The composition of the diluting media was adjusted to impose a fixed inward HCO_3^- gradient of 3:1 while maintaining $(\text{K}^+)_o = (\text{K}^+)_i$ in the presence of valinomycin. The net change in intravesicular content of ^{22}Na was then assayed over the 3-s period immediately following the imposition of these ion gradients. As shown in Fig. 2, when a 9:1 outward Na^+ gradient was imposed, the 3:1 inward HCO_3^- gradient induced the net influx of Na^+ . If $n = 2$, the cotransport system would have been at equilibrium and no net flux would have been observed under these conditions. The fact that the 3:1 inward HCO_3^- gradient could overcome the 9:1 outward Na^+ gradient as a driving force indicates that n must be > 2 . The 3:1 inward HCO_3^- gradient even caused net Na^+ influx when outward Na^+ gradients of 12.5:1 or 17.5:1 were imposed, which would have brought the system to equilibrium if n were equal to 2.3 or 2.6, respectively. These results indicate that n must be > 2.6 . In contrast, when a 46.7:1 outward Na^+ gradient was imposed, which would have brought the system to equilibrium if $n = 3.5$, net Na^+ efflux was observed. Thus, this experiment indicates that the value for n must be between 2.6 and 3.5. Indeed, when outward Na^+ gradients of 24.2:1 and 33.5:1 were

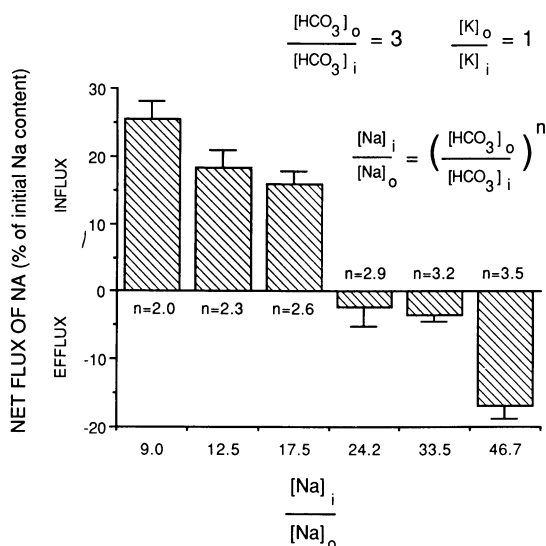


Figure 2. Net Na^+ flux in the presence of a 3:1 inward HCO_3^- gradient and varying outward Na^+ gradients. Basolateral membrane vesicles were pre-equilibrated for 120 min at 20°C in a medium that consisted of 1 mM ^{22}Na -gluconate, 52 mM TMA-gluconate, 42 mM Hepes, 23 mM TMA-OH, 38 mM mannitol, 14 mM Mes, 21 mM K-HCO₃, and 50 mM K-gluconate, pH 7.06, gassed with 90% N_2 , 10% CO_2 . Intravesicular content of ^{22}Na was then measured before and 3 s after 1:9, 1:12.5, 1:17.5, 1:24.2, 1:33.5, or 1:46.7 dilution and incubation of the vesicles in a medium of final composition 53 mM TMA-gluconate, 42 mM Hepes, 23 mM TMA-OH, 50 mM mannitol, 2 mM Mes, 63 mM K-HCO₃, and 8 mM K-gluconate, pH 7.50, gassed with 90% N_2 , 10% CO_2 . Values shown for net flux represent means \pm SE for three separate experiments performed in triplicate on three different membrane preparations. For each imposed Na^+ gradient, the value of n is shown that would satisfy the illustrated form of Eq. 4, which applies when $(\text{K}^+)_o = (\text{K}^+)_i$.

imposed, corresponding to n values of 2.9 and 3.2 respectively, no significant net Na^+ flux was observed, again consistent with a cotransport stoichiometry of 3 HCO_3^- : Na^+ .

In the experimental approach illustrated in Fig. 3, membrane vesicles were preloaded with ^{22}Na and then diluted 10-fold so that a 10:1 outward Na^+ gradient was imposed, and the composition of the diluting media was adjusted to impose different inward K^+ gradients in the presence of valinomycin while maintaining $(\text{HCO}_3^-)_o = (\text{HCO}_3^-)_i$. The net change in intravesicular content of ^{22}Na was then assayed over the 3-s period immediately following the imposition of these ion gradients. Shown at the far right in Fig. 3, imposing a 10:1 inward K^+ gradient induced the net influx of Na^+ against its own 10:1 outward gradient. According to Eq. 4, if $n = 2$, the cotransport system would have been at equilibrium and no net flux would have been observed under these conditions. The fact that the 10:1 inward K^+ gradient could overcome the 10:1 outward Na^+ gradient as a driving force indicates that n must be > 2 . When a 4.2:1 inward K^+ gradient was imposed, net Na^+ influx was again observed, indicating that n must be > 2.6 . In contrast, when a 2.2:1 inward K^+ gradient was imposed, which would have brought the system to equilibrium if $n = 4.0$, net Na^+ efflux was observed. Thus, this experiment indicates that the value for n must be between 2.6 and 4.0. Indeed, when a 3.2:1 inward K^+ gradient was imposed no

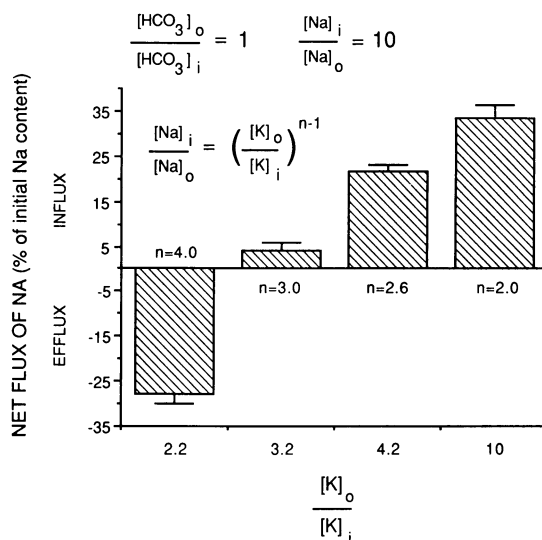


Figure 3. Net Na^+ flux in the presence of a 10:1 outward Na^+ gradient and varying inward K^+ gradients. Basolateral membrane vesicles were pre-equilibrated for 120 min at 20°C in a medium that consisted of 1 mM ^{22}Na -gluconate, 66 mM TMA-gluconate, 52 mM mannitol, 42 mM Hepes, 21 mM TMA-OH, pH 7.50, gassed with 90% N_2 , 10% CO_2 , and that contained in addition either 46 mM choline-HCO₃ and 11 mM K-HCO₃, 30 mM choline-HCO₃ and 27 mM K-HCO₃, 22 mM choline-HCO₃ and 35 mM K-HCO₃, or 5 mM choline-HCO₃ and 52 mM K-HCO₃, corresponding to inward K^+ gradients of 10, 4.2, 3.2, or 2.2:1, respectively. Intravesicular content of ^{22}Na was then measured before and 3 s after 1:10 dilution and incubation of the vesicles in a medium of final composition 7 mM TMA-gluconate, 42 mM Hepes, 21 mM TMA-OH, 5 mM choline-HCO₃, 52 mM K-HCO₃, and 60 mM K-gluconate, pH 7.50, gassed with 90% N_2 , 10% CO_2 . Values shown for net flux represent means \pm SE for three separate experiments performed in triplicate on three different membrane preparations. For each imposed K^+ gradient, the value of n is shown that would satisfy the illustrated form of Eq. 4, which applies when $(\text{HCO}_3^-)_o = (\text{HCO}_3^-)_i$.

significant net Na⁺ flux was observed, consistent with a cotransport stoichiometry of 3 HCO₃⁻:Na⁺.

A similar approach was used in the next experiment (Fig. 4) to define the stoichiometry of cotransport more precisely. In this case, vesicles were preloaded with ²²Na, and then different outward Na⁺ gradients were imposed by diluting 1 vol of Na⁺-loaded vesicles in different volumes of Na⁺-free media. The composition of the diluting media was adjusted to impose a fixed inward K⁺ gradient of 3:1 in the presence of valinomycin while maintaining (HCO₃⁻)_o = (HCO₃⁻)_i. The net change in intravesicular content of ²²Na was then assayed over the 3-s period immediately following the imposition of these ion gradients. As shown in Fig. 4, the 3:1 inward K⁺ gradient induced the net influx of Na⁺ in the presence of outward Na⁺ gradients of 3:1, 4.2:1 or 5.8:1, which would have brought the system to equilibrium if *n* = 2.0, 2.3, or 2.6, respectively. This clearly indicates that *n* must be > 2.6. In contrast, when a 15.5:1 outward Na⁺ gradient was imposed, which would have brought the system to

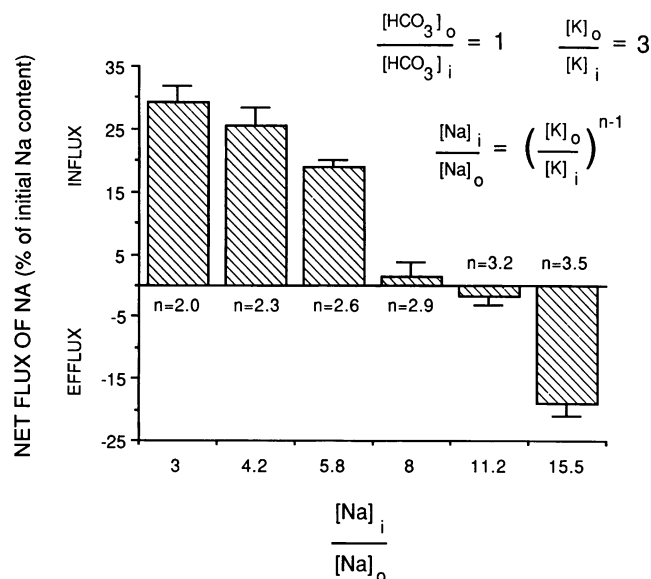


Figure 4. Net Na⁺ flux in the presence of 3:1 inward K⁺ gradient and varying outward Na⁺ gradients. Basolateral membrane vesicles were preequilibrated for 120 min at 20°C in a medium that consisted of 1 mM ²²Na-gluconate, 66 mM TMA-gluconate, 52 mM mannitol, 42 mM Hepes, 21 mM TMA-OH, 27 mM choline-HCO₃, and 30 mM K-HCO₃, pH 7.50, gassed with 90% N₂, 10% CO₂. Intravesicular content of ²²Na was then measured before and 3 s after 1:3, 1:4.2, 1:5.8, 1:8, 1:11.2, or 1:15.5 dilution and incubation of the vesicles in a medium of final composition 53 mM TMA-gluconate, 42 mM Hepes, 21 mM TMA-OH, pH 7.50, gassed with 90% N₂, 10% CO₂, and that contained in addition either 24 mM TMA-gluconate, 43 mM K-gluconate, 47 mM K-HCO₃, and 10 mM choline-HCO₃; 28 mM TMA-gluconate, 39 mM K-gluconate, 51 mM K-HCO₃, and 6 mM choline-HCO₃; 23 mM TMA-gluconate, 44 mM K-gluconate, 46 mM K-HCO₃, and 11 mM choline-HCO₃; 21 mM TMA-gluconate, 46 mM K-gluconate, 44 mM K-HCO₃, and 13 mM choline-HCO₃; 19 mM TMA-gluconate, 48 mM K-gluconate, 42 mM K-HCO₃, and 15 mM choline-HCO₃; or 18 mM TMA-gluconate, 49 mM K-gluconate, 41 mM K-HCO₃, and 16 mM choline-HCO₃; respectively. Values shown for net flux represent means ± SE for three separate experiments performed in triplicate on three different membrane preparations. For each imposed Na⁺ gradient, the value of *n* is shown that would satisfy the illustrated form of Eq. 4, which applies when (HCO₃⁻)_o = (HCO₃⁻)_i.

equilibrium if *n* = 3.5, net Na⁺ efflux was observed. Thus, this experiment indicates that the value for *n* must be between 2.6 and 3.5. Indeed, when outward Na⁺ gradients of 8:1 and 11.2:1 were imposed, corresponding to *n* values of 2.9 and 3.2 respectively, no significant net Na⁺ flux was observed, again consistent with a cotransport stoichiometry of 3 HCO₃⁻:Na⁺.

It should be emphasized that although all of the above results are consistent with a cotransport stoichiometry of 3 HCO₃⁻ per Na⁺, they are equally consistent with any transport process in which there is the net transfer of three equivalents of base, one Na⁺, and two negative charges per transport event. In our experimental protocols using a CO₂/HCO₃⁻ buffer system, whenever inward HCO₃⁻ gradients were imposed, inward OH⁻ and outward H⁺ gradients of equal magnitude were also imposed. Thus, for example, the cotransport of Na⁺ with 3 HCO₃⁻, the cotransport of Na⁺ with 2 HCO₃⁻ and 1 OH⁻, and the cotransport of Na⁺ with 2 HCO₃⁻ in exchange for 1 H⁺ are thermodynamically equivalent processes that cannot be distinguished by the approach we have utilized. Similarly, our approach will not discriminate between the cotransport Na⁺ with 3 HCO₃⁻ and the cotransport of Na⁺ with 1 HCO₃⁻ and 1 CO₃²⁻.

In conclusion, we have demonstrated that the stoichiometry of the Na⁺-HCO₃⁻ cotransport system in rabbit renal basolateral membrane vesicles is 3 HCO₃⁻:Na⁺ or a thermodynamically equivalent process. With such a stoichiometry, the inside-negative membrane potential is sufficient to drive HCO₃⁻ exit against the inward concentration gradients of HCO₃⁻ and Na⁺ that are present across the basolateral membrane of the intact proximal tubule cell under physiologic conditions (5, 13).

Acknowledgments

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