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SOME OBSERVATIONS ON THE METABOLISM OF D-GALACTOSE IN NORMAL MAN

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For many years clinical interest in D-galactose centered about the usefulness of galactose tolerance tests in the diagnosis of impaired liver function. Several investigators, employing an intravenous galactose tolerance test, have reported their results (1-3). The monograph by Stenstam (4) is perhaps the most extensive analysis of both the oral and intravenous tolerance test yet reported. In all of these studies the parameter for measurement of galactose metabolism has been the disappearance rate of the sugar from blood or the urinary excretion of the sugar for an arbitrary time interval.

More recently research in galactose metabolism has had as its focal point the disease, congenital galactosemia. Kalckar and Maxwell, and their associates (5-7) have contributed to the delineation of the pathways of galactose metabolism and have pinpointed the enzymatic defect in galactosemia (8, 9). Other investigators have reported that certain steroids as well as menthol are able to prevent the depression of galactose oxidation in liver slices and homogenates caused by the metabolism of ethanol and aldehydes (10-12). The administration of progesterone (13) and menthol (12) has enhanced the ability of galactosemic subjects to oxidize galactose to CO₂.

Because of the latter experiments it became important to delineate the physiologic disposition and metabolic fate of C¹⁴-galactose in normal subjects. This report presents the results of experiments performed for this purpose. Observations have been made of the effect of progesterone, menthol, and ethanol on galactose-1-C¹⁴ oxidation.

METHODS

Sixteen intravenous infusions of galactose-1-C¹⁴ were carried out in 6 normal male volunteer subjects whose ages ranged from 18 to 22. These subjects were maintained on a 300 g carbohydrate diet and had normal glucose tolerance as determined by the method of Amatuzio, Stutzman, Vanderbilt and Nesbitt (14). All experi-

ments were performed after an overnight fast which was continued during the 5-hour period of study.

All of the above subjects received 5 μc of galactose-1-C¹⁴ either as a tracer of 1.05 mg or mixed with either 10 or 20 g of unlabeled galactose present in 20 per cent solution. In two studies 50 mg per day i.m. progesterone and in another experiment 750 mg per day p.o. menthol were administered for 5 and 3 days, respectively, prior to the isotope infusion. When ethanol was given, 10 or 20 ml of absolute ethanol was diluted with an equal volume of water and ingested, p.o., 5 minutes prior to the radioactive galactose infusion. At various time intervals after the infusion, expired air for C¹⁴O₂ analysis was collected in Douglas bags and blood was drawn for galactose and glucose analysis. Urine was collected for 24 hours for determination of radioactivity, and in some experiments in which large quantities of galactose were given, assays were performed for reducing substance and glucose oxidase-reacting material.

Galactose-1-C¹⁴ (specific activity 4.72 μc per mg) was purchased from Dr. H. Isbell of the National Bureau of Standards and prepared for intravenous use as a solution of 1 μc per ml of normal saline by the radiopharmacy of the National Institutes of Health. Unlabeled galactose was purchased from the Pfanstiehl Co. and prepared for i.v. use as a 20 per cent solution by the NIH pharmacy. These solutions were sterile and pyrogen-free. USP absolute ethanol was used. Progesterone in oil was the commercial preparation of Eli Lilly and Co. L-menthol was given orally as a 25 per cent peanut oil solution in gelatin capsules containing 250 mg of the drug. The Na pyruvate used was purchased from Schwartz Biochemical Co. and prepared for use as a 10 per cent solution in normal saline.

Carbon dioxide was assayed by the method of Fredrickson and Ono (15) and the C¹⁴O₂ in Hyamine assayed in a Packard Tri-Carb liquid scintillation spectrometer counting at 54 per cent efficiency. Blood glucose in filtrates prepared by the Somogyi procedure and urine glucose were assayed with glucose oxidase.¹ Galactose was estimated as the difference between total reducing substance by the Nelson method (16) and glucose oxidase-positive material. Blood was assayed for radioactivity as described previously (17). Glucose was isolated from blood by the method of Blair and Segal (18). Radioactive galactose in blood was estimated by the isotope dilution technic as follows. To a 1:10 Somogyi blood

¹ Glucostat reagent purchased from Worthington Biochemical Co., Freehold, N. J.

filtrate made from 20 ml whole blood was added a known amount (approximately 50 mg) of galactose. The filtrate was evaporated to 3 ml. To this was added an equal volume of concentrated nitric acid. This was heated for 90 minutes in a boiling water bath and then placed overnight in a refrigerator at 4° C. The mucic acid crystallized out of solution and was recrystallized from hot water. Both gluconate and mucic acid were combusted to CO₂ by the method of Van Slyke, Plazin and Weisiger (19). The BaCO₃ thus obtained was treated with acid to liberate CO₂ which was trapped in Hyamine and counted by the scintillation technic (20). Urine-C¹⁴ was assayed as previously described (21).

Two laboratory personnel, one male and one female, ages 33 and 35, respectively, donated blood from which leukocytes were prepared by the method of Skoog and Beck (22). The incubation procedure was similar to that already reported (23). Human liver for *in vitro* study was obtained at surgery from a 29-year old male

with metastatic islet cell carcinoma. Tumor was present in the liver but none was observed macroscopically in the segments used to prepare the tissue slices for the *in vitro* study.

RESULTS

The effect of increasing the quantity of sugar injected on oxidation of galactose-1-C¹⁴ to C¹⁴O₂. The radioactive sugar was metabolized to labeled CO₂ as shown for a tracer amount in Figure 1. The control curve in the upper part of Figure 1 is similar to that observed after glucose-1-C¹⁴ administration (24). Table I reveals the per cent of the administered C¹⁴ found in expired air at hourly intervals after injection of the isotope. When the amount of injected galactose was about 1 mg, 31 to 35 per cent of the C¹⁴ dose was in ex-

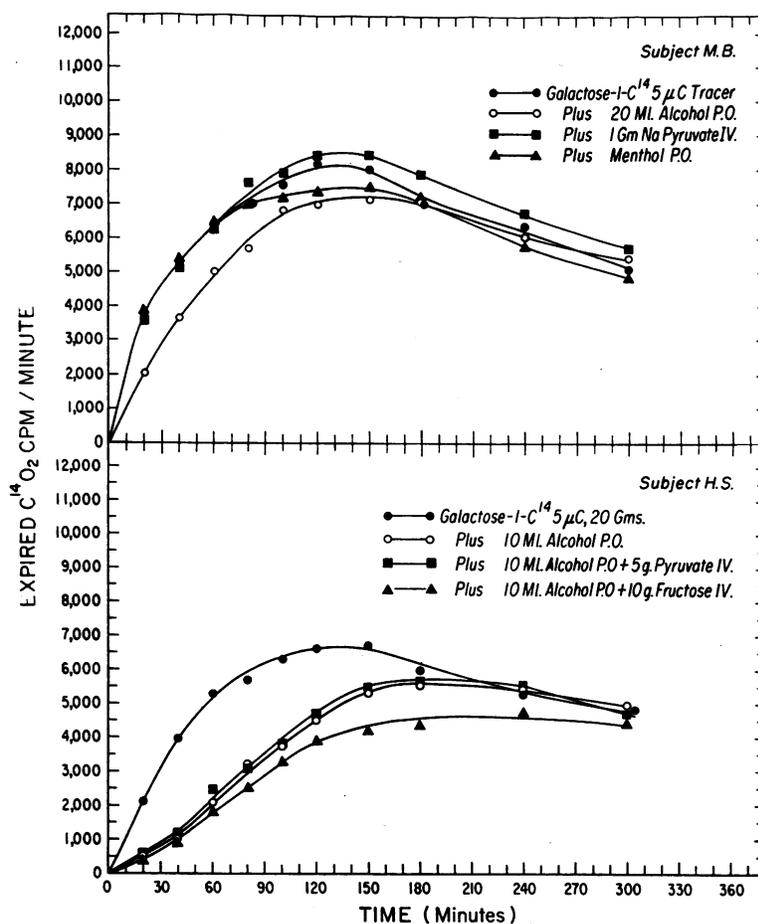


FIG. 1. UPPER PORTION IS THE EXPIRED C¹⁴O₂ AFTER I.V. ADMINISTRATION OF A TRACER QUANTITY OF GALACTOSE-1-C¹⁴ IN A NORMAL SUBJECT. Alcohol refers to ethanol. Lower portion is the C¹⁴O₂ excretion curves after administration of 20 g galactose containing the same amount of galactose-1-C¹⁴ as given in the tracer experiments.

TABLE I
The extent of oxidation of galactose-1-C¹⁴ after intravenous injection into normal subjects

Subj.	Amt. of sugar	Exptal. additions	CO ₂ excret.	Administered C ¹⁴ in expired air				
				Minutes after injection				
	g		mmoles/min	60	120	180	240	300
S.K.	0.00106	None	10.8	3.86	10.17	17.60	25.08	31.71
	0.00106	Progesterone	10.8	3.84	10.53	18.23	25.75	32.42
J.F.	0.00106	None	12.2	4.31	12.54	21.25	28.77	35.23
	0.00106	Progesterone	12.7	5.49	14.75	24.41	32.85	39.85
M.B.	0.00106	None	9.9	4.24	11.87	20.09	27.19	33.07
	0.00106	Menthol	10.0	4.24	11.53	19.14	25.80	31.30
	0.00106	Pyruvate, 1 g	10.6	4.24	12.13	20.69	28.22	34.58
	0.00106	Ethanol, 20 ml	9.6	2.80	9.19	16.57	23.50	29.43
D.S.	10.0	None	10.3	2.01	7.20	14.40	22.09	29.11
	10.0	Ethanol, 10 ml	10.5	1.02	5.23	11.62	18.45	24.93
	Glucose-1-C ¹⁴ 0.0018	None	10.6	4.02	12.50	21.61	29.31	35.56
	0.0018	Ethanol, 10 ml	11.1	4.69	14.01	23.42	31.58	38.54
D.P.	20.0	None	11.7	1.85	6.66	12.75	19.10	25.24
	20.0	Ethanol, 20 ml	13.0	0.77	3.19	7.93	14.03	20.11
H.S.	20.0	None	10.1	3.08	9.49	16.37	22.51	27.78
	20.0	Ethanol, 10 ml	10.7	0.94	4.47	9.98	15.81	21.27
	20.0	Ethanol, 10 ml						
		Fructose, 10 g	9.7	0.81	3.84	8.30	13.14	17.84
	20.0	Ethanol, 10 ml Pyruvate, 5 g	10.5	1.02	4.73	10.44	16.41	21.83

pired CO₂ within 5 hours. This corresponds quite closely with the amount of C¹⁴O₂ excreted after glucose-1-C¹⁴ injection as reported previously (24) and shown also in Subject D.S., Table I. The injection of 10 and 20 g sugar containing an identical amount of radioactivity resulted in a slight reduction of the per cent of the injected C¹⁴ excreted in expired air, as shown in Table I. Subject D.S., given 10 g, excreted 29 per cent, and Subjects D.P. and H.S., given 20 g, excreted 25 and 28 per cent of the radioactivity, respectively, in 5 hours.

The effect of increasing the amount of injected sugar is most striking when one compares the per cent C¹⁴ excreted at 1 hour with the values for the tracer studies. The lower portion of Figure 1 shows the C¹⁴O₂ excretion curve after a 20 g infusion in Subject H.S. The initial rate of C¹⁴O₂ production is about 40 per cent slower than that seen for the tracer study in Figure 1. With time, however, the C¹⁴O₂ excretion after a load approaches that seen after a tracer quantity. Considering the fact that the specific activity of the injected galactose is lowered 10- and 20-thousand-

fold in the load experiments, it appears that the system has a large capacity for metabolizing galactose. The fact that the C¹⁴O₂ excretion is somewhat lower after a load suggests that the system is beginning to be saturated, for it would be expected that the per cent of substrate metabolized remains constant with increasing concentration until the point of saturation where the value would then decrease. The effect on expired C¹⁴O₂ observed with increasing quantity of injected galactose is in marked contrast to that observed with D-ribose, where a marked decrease in oxidation occurred upon increasing the sugar given from a tracer to 20 g (17).

Effect of progesterone and menthol on galactose oxidation. It has been reported that progesterone administration to three galactosemic children and menthol administration in two others enhanced their ability to oxidize a tracer quantity of labeled galactose (12, 13). In these experiments, during control periods, virtually no C¹⁴O₂ was produced, whereas after the administration of these drugs, from 7 to 17 per cent of the C¹⁴ injected appeared in CO₂ in 5 hours. Three studies, two

with progesterone and one with menthol, have now been performed in normal subjects, and the results are shown in Table I and Figure 1. A problem presented itself in these studies. Whereas, in the galactosemic children in whom the capacity for metabolizing galactose to CO_2 is essentially nil and the effects of a drug stimulating oxidation may be easily seen, in the normal there is a baseline of rapid and extensive metabolism which is probably maximal. Fluctuations in this baseline could make interpretation of the drug studies difficult. In Subject S.K. the per cent of the C^{14} given that appeared in expired air was virtually identical before and after progesterone administration. Subject J.F. showed a slight increase after the drug. Subject M.B., whose post-menthol excretion curve is seen in Figure 1, had no stimulatory effect of the drug. The results indicate that these drugs do not have an effect on galactose metabolism in a normal subject.

Caution must be exercised, however, in comparing the effect of these drugs in post-pubertal normal subjects and children. Because of the problems involved, we have not performed these experiments with radioisotopes in normal children.

Effect of ethyl alcohol on galactose oxidation. Stenstam (4) has demonstrated very clearly in the human that ethyl alcohol impairs the clearance of administered galactose from blood. Several reports have appeared which demonstrate *in vitro* that ethanol inhibits galactose metabolism and that this effect is due to changes in intracellular reduced diphosphopyridine nucleotide (DPNH) (11, 25, 26).

The present studies reveal that small amounts of ethanol may markedly inhibit the metabolism of galactose-1- C^{14} to C^{14}O_2 . These results are tabulated in Table I. Figure 1 demonstrates this inhibition in a tracer study in Subject M.B. The potency of the effect is brought out here, since this reveals the inhibition of metabolism of 1 mg in a system that has a capacity for metabolizing many thousands of milligrams. The effect of ethanol was greater in the loading experiments with 10 and 20 g. Figure 1 also shows the marked inhibition of appearance of C^{14}O_2 in a study with 20 g of the sugar. This effect of ethanol is not seen when C^{14} glucose is injected, as is shown in Table I, Subject D.S.

Smith and Newman (27) have demonstrated

in the rat that injected pyruvate decreases the high levels of DPNH generated in liver during ethanol administration by being reduced to lactate with DPNH as a cofactor. Two experiments were therefore performed on the effect of pyruvate on galactose-1- C^{14} oxidation. In the first (Figure 1), 1 g of pyruvate was given i.v. in the absence of ethanol to see whether the normal metabolism could be stimulated. Little or no effect was observed. In the second, shown in the lower section of Figure 1, an attempt was made to reverse the effect of 10 ml of ethanol by injecting 5 g of sodium pyruvate. This amount of pyruvate failed to overcome the inhibition. Because the principal fate of pyruvate in man may be oxidation rather than the conversion to lactate, which would be needed for alteration of DPNH levels, a pyruvate-generating system, involving fructose metabolism, was employed (28). Figure 1 also demonstrates that 10 g of fructose i.v. was also unable to overcome the ethanol effect.

The disappearance of C^{14} and galactose from blood and the conversion of galactose to glucose. Figure 2 demonstrates the change with time in whole blood C^{14} after a tracer dose of galactose-1- C^{14} was given to Subjects M.B. and J.F. The inset shows the decrease in chemically determined galactose in blood of M.B. after a 5 g infusion. Whereas there was no detectable galactose in blood 15 minutes after the 5 g infusion, at the same time and for many minutes thereafter, much C^{14} was present in the tracer experiment. The rate of decrease of C^{14} was very much slower than that of galactose itself.

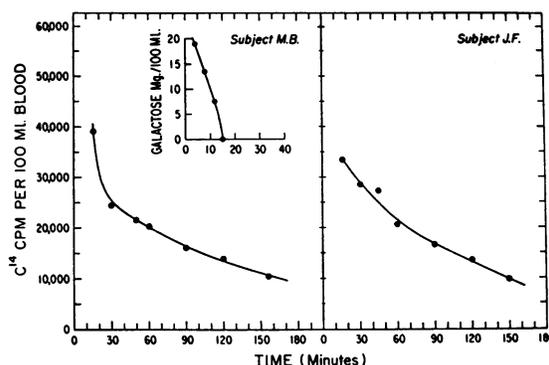


FIG. 2. THE CHANGE IN BLOOD C^{14} WITH TIME AFTER LABELED GALACTOSE INJECTION. The inset shows the disappearance from blood of chemically determined galactose after a 5 g injection.

TABLE II
Distribution of C^{14} in blood after galactose-1- C^{14}
and glucose-1- C^{14} administration

Sugar injected and subject	Time after inject.	C^{14} in blood		
		Total C^{14}	Galactose C^{14}	Glucose C^{14} *
	min	dpm/100 ml	dpm/100 ml	dpm/100 ml
Galactose-1- C^{14} M.B.	30	48,400	2,723	18,485
	121	21,000	223	7,305
J.F.	15	43,400	3,402	12,904
	96	20,600	657	5,980
Glucose-1- C^{14} D.S.	30	35,600		29,680
	90	22,380		10,250

* Corrected for adsorption of radioactive galactose (18).

The latter fact suggested that metabolites of galactose were present in blood. Since the pathway of galactose metabolism involves conversion to glucose, two experiments were carried out to ascertain how much of C^{14} in blood after a tracer injection was in glucose or galactose. The galactose- C^{14} was isolated as mucic acid by a carrier-dilution technic, and glucose as gluconate. Table II contains the results of these experiments. In J.F. 15 minutes, and in M.B. 30 minutes, after the injection, less than 10 per cent of the blood- C^{14} is galactose. At the latter time intervals shown, galactose- C^{14} is even a smaller fraction of the total C^{14} . On the other hand, a substantial amount of the C^{14} is present as glucose. The nature of the other C^{14} compound(s) is unknown. For comparison, results are shown in the table for the amount of C^{14} in blood glucose after glucose-1- C^{14} injection. With time a substantial amount of non-glucose- C^{14} appears in blood, paralleling the results in the galactose experiments.

It is possible to estimate the fraction of injected galactose- C^{14} circulating as glucose by determining the total C^{14} in the glucose compartments. This may be calculated if the specific activity of blood glucose and the size of the glucose pool are known. For the 30-minute point of Subject M.B. the specific activity of blood glucose was 52,700 dpm per mmole. Estimating the glucose pools to be 100 mmoles (24), it may be calculated that about 5 million dpm of the 11 million injected are in body glucose pools. This result is consistent with the known pathway of galactose metabolism in which glucose phosphates are intermediates in the conversion of galactose to CO_2 .

In Figure 3 are shown data of the effect of

ethyl alcohol on galactose disappearance from blood after a 20 g infusion. As has been demonstrated previously by Stenstam (4), alcohol slows this disappearance and this was not corrected by infusion of 5 g sodium pyruvate, paralleling the results on $C^{14}O_2$ described previously in the section on $C^{14}O_2$ excretion. Figure 3 also shows a rise in blood glucose levels after galactose infusion and this is consistent with the data above on the conversion of galactose to glucose. It should be noted that when alcohol is ingested there is little or no increase in blood glucose, as would be expected if alcohol inhibited the conversion of galactose to glucose.

In certain normal individuals (4, 29) and in galactosemic infants (30), galactose infusion may cause hypoglycemia. Pozza, Galansino, Hoffeld and Foa (31) have demonstrated that galactose infusion causes increased insulin release from the pancreas. Hyperreactivity to insulin may explain the resultant hypoglycemia in these people. The alternative explanation of Sidbury (32) and Ginsburg and Neufeld (33) that phosphoglucomutase is inhibited by accumulated galactose-1-phosphate, thereby slowing glycogen breakdown to glucose, also should be entertained.

Urinary excretion of C^{14} , reducing substance and glucose. Table III presents data on the appearance of C^{14} in urine. About 3 per cent of the injected C^{14} was excreted by this route in the tracer studies. In the one study with menthol (Subject M.B.) significantly less C^{14} appeared in urine. With injection of 20 g, 5 to 8 per cent of the C^{14} originally given was excreted. Alcohol in-

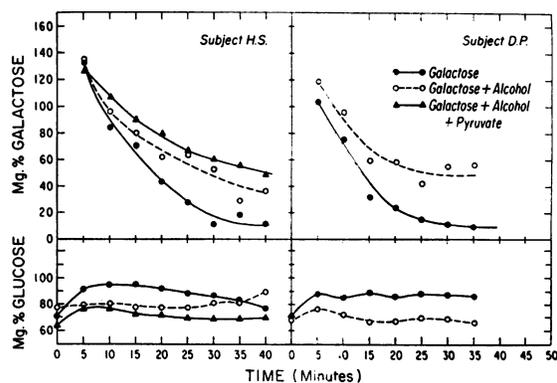


FIG. 3. DISAPPEARANCE OF CHEMICALLY DETERMINED GALACTOSE FROM BLOOD AFTER INFUSION OF 20 G AND THE CONCOMITANT CHANGE IN BLOOD GLUCOSE.

TABLE III

Urinary excretion of C¹⁴, reducing substance and glucose oxidase-reacting substance after galactose administration

Subj.	Amt. of galactose inj.	Drug	Hours:	Dose excreted					
				C ¹⁴		Reducing substances		Glucose oxidase reactant	
				0-5	5-24	0-5	5-24	0-5	5-24
	<i>g</i>			<i>%</i>		<i>%</i>		<i>%</i>	
M.B.	0.00106	None		2.7	0.7				
	0.00106	Menthol		0.3	0.6				
	0.00106	Alcohol, 10 ml		3.2	0.5				
	0.00106	Pyruvate, 1 g			3.6*				
J.F.	0.00106	None		1.6	1.6				
	0.00106	Progesterone		1.8	0.8				
D.S.	20.0	None		4.5	0.7	7.8	8.3	0.42	0.55
	20.0	Alcohol, 10 ml		8.6	0.5	13.3	3.1	0.20	0.34
D.P.	20.0	None		7.1	0.9	11.2	3.6	0.31	2.76
	20.0	Alcohol, 20 ml		17.7	0.7	17.6	1.4	0.53	0.96

* Total 24-hour excretion.

gestion essentially doubled the 5-hour C¹⁴ excretion. Reducing substance was assayed in urine of the subjects receiving 20 g in order to see whether the C¹⁴ excreted corresponded to copper-reducing sugar. In all but one urine specimen, reducing substance expressed as galactose was greater than the C¹⁴. This led to the suspicion that urinary glucose excretion was elevated and that this might account for the discrepancy between reducing substance and C¹⁴. The glucose excretion, however (Table III), was not greater than that reported by Froesch and Renold (34) for normal urine, and did not account for the difference

between C¹⁴ and reducing substance. Froesch and Renold have shown that as much as 1 g of nonspecific reducing substance expressed as glucose was excreted in 24-hour urine specimens. Since the reducing power of galactose is less than glucose, if the reducing substance is expressed as galactose it would represent well over 1 g of the sugar or over 5 per cent of the injected 20 g. Thus, the difference between the C¹⁴ and galactose expressed as reducing substance could be explained on the basis of nonspecific reducing substance.

In vitro effect of various drugs on galactose oxi-

TABLE IV

*In vitro effect of ethanol, menthol, progesterone, and pyruvate on galactose-1-C¹⁴ oxidation by human leukocytes and liver**

	cpm/10 ⁸ leukocytes or/100 mg. liver†											
	Control	Ethanol 10 ⁻² M									Pyruvate	
		Menthol				Progesterone				8 × 10 ⁻³	1.6 × 10 ⁻² M	3.2 × 10 ⁻² M
	10 ⁻⁴	10 ⁻⁵	10 ⁻⁶	10 ⁻⁸ M	10 ⁻⁴	10 ⁻⁶	10 ⁻⁸	10 ⁻¹⁰ M				
WBC												
Exp. 1; Subj. A.B.	148	115		112	107		109	100				
Exp. 2; Subj. A.B.	136	103					114	98	113			
Exp. 3; Subj. S.S.	167	126	99				97			250	385	445
Liver												
Exp. 1; Subj. C.C.	2,736	1,279	2,282	1,340			766					

* Liver slices were prepared with a Stadie-Riggs microtome and placed in modified Warburg flasks containing 2 ml of Krebs-Ringer bicarbonate buffer and 0.24 μc of radioactive galactose. Incubation procedure was similar to that described for the leukocytes (23).

† Leukocyte values are averages of duplicate, and liver the averages of triplicate determinations.

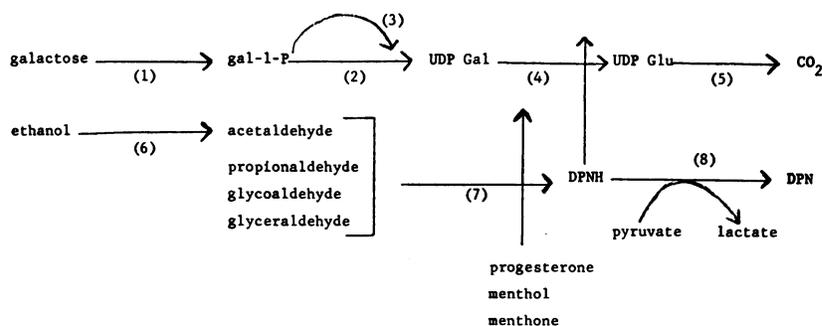


FIG. 4. SCHEMATIC REPRESENTATION OF THE PATHWAY AND FACTORS AFFECTING GALACTOSE METABOLISM. The numbers represent reactions involving the following enzymes: 1, galactokinase; 2, galactose-1-P uridyl transferase; 3, uridinediphosphate galactose (UDPGal) pyrophosphorylase; 4, UDPGal-4-epimerase; 5, uridinediphosphate glucose (UDPGlu) pyrophosphorylase and subsequent enzymes from glucose-1-P to CO₂; 6, alcohol dehydrogenase; 7, aldehyde dehydrogenase; 8, lactic dehydrogenase.

dation by human leukocytes and liver. The inhibition of galactose oxidation in animal liver and intestine by ethanol and certain aldehydes and the reversal of this effect by menthol and progesterone have been reported (10-12). Experiments were performed on human leukocytes to ascertain whether these effects were obtained in human tissues. The data in Table IV demonstrate that ethanol was able to depress galactose oxidation in leukocytes but that menthol and progesterone in a wide range of concentration did not restore the galactose oxidation to normal. On the other hand, pyruvate not only was able to reverse the inhibition caused by ethanol but also caused a stimulation of galactose oxidation above the baseline values. This is in contrast to the lack of effect of pyruvate in the *in vivo* studies described above. Other experiments with normal hemolysates have shown the inability of menthol and progesterone to reverse the depression of galactose metabolism caused by aldehydic substances (35); but when human liver was employed, 10⁻⁴ M menthol tended to negate the ethanol effect. Progesterone at the same concentration appeared to have no such effect.

DISCUSSION

D-galactose is an important sugar in human nutrition, especially in the young, whose principal food is milk. Milk contains about 5 g per 100 ml of lactose, which is a disaccharide of glucose and galactose (36). It is not surprising, therefore, to

find that normal subjects have the capacity to metabolize large quantities of injected galactose.

The pathway of galactose metabolism has been delineated by Caputto and co-workers and Leloir (37, 38) and by Kalckar and Maxwell and associates (5, 6). This is schematically shown in Figure 4. Galactose is phosphorylated to form galactose-1-phosphate (gal-1-P) which is then converted to UDPGal by the enzyme gal-1-P uridyl transferase (reaction 2). This is the deficient enzyme in galactosemia. By an alternate route found in yeast (5), mung bean seedlings (39), and in liver (40) (reaction 3), gal-1-P may be transformed to UDPGal. The physiologic role of this pathway in man remains to be assessed. UDPGal is epimerized by UDPGal-4-epimerase to UDP glucose (reaction 4). The latter may undergo pyrophosphorylytic cleavage to glucose-1-phosphate which then enters the usual pathway of glucose oxidation to CO₂. The present studies show that, in man, galactose is extensively converted to glucose.

Maxwell (41) has described the properties of the UDPGal-4-epimerase enzyme. One of these is marked inhibition by DPNH. DPN is an essential cofactor for the epimerase reaction. It has now become apparent that the level of DPNH in tissue exerts a marked controlling effect on galactose metabolism. Figure 4 summarizes schematically the factors regulating galactose metabolism (10-12, 25, 26). Ethanol via conversion to

acetaldehyde, and various aldehydes via conversion to their respective acids by aldehyde dehydrogenase, generate DPNH (reactions 6 and 7). The DPNH inhibits the epimerization of UDPGal which interferes with galactose conversion to glucose. Progesterone, menthol and menthone in the *in vitro* studies inhibit aldehyde dehydrogenase (reaction 7), thereby preventing the rise in DPNH levels. The effect of ethanol and aldehydes may be reversed by pyruvate conversion to lactate which involves DPNH oxidation to DPN (reaction 8). This takes the "brake" off the epimerase reaction. The importance of the epimerase reaction in the control of galactose metabolism even in the presence of limited amounts of uridyl transferase enzyme (reaction 2) has recently been stressed (42).

The explanation for the lack of an effect in white cells of progesterone and menthol is not readily apparent. Menthol and progesterone are believed to act by affecting the activity of aldehyde dehydrogenase. Maxwell has purified human liver aldehyde dehydrogenase and has found it to be less sensitive to these drugs than are the corresponding enzymes prepared from several species (43). This may be the explanation for an incomplete effect of menthol and the absence of a progesterone effect at the concentrations used in the experiments with human liver. The lack of an effect of the drugs in the white cell could be related to their permeability into the cell. The insensitivity of red cell aldehyde dehydrogenase to these substances (35) tends to suggest, however, that the enzyme present in peripheral blood cells may have characteristics different from the corresponding enzyme present in liver.

Although the role of progesterone and menthol in galactose metabolism in animal tissues *in vitro* has been elucidated, the mechanism of action of these drugs to stimulate galactose metabolism in subjects with galactosemia remains obscure. The explanation of the latter effect becomes more puzzling in view of the present findings that these drugs do not stimulate galactose metabolism in normal subjects. This raises the question of whether it is mere coincidence that progesterone and menthol *in vitro* act primarily in a system depressed by aldehyde oxidation and *in vivo* only in subjects whose metabolism is also depressed.

The knowledge that ethanol ingestion inhibits galactose metabolism has some clinical application. Schwarz has found the accumulation of galactose-1-phosphate in cord blood of a newborn galactosemic baby (44) and, along with others (45), has postulated that there may be *in utero* damage to such a child by transplacental galactose. It might be recommended that when a mother of a known galactosemic child becomes pregnant, in addition to institution of a galactose-free diet, cessation of any ethanol ingestion should be carried out as well.

SUMMARY

Various quantities of galactose-1-C¹⁴, when given intravenously to normal human subjects, are rapidly and extensively metabolized to C¹⁴O₂. This metabolism is markedly slowed by ingestion of ethanol but is not affected by administration of progesterone, menthol or pyruvate. White blood cell metabolism of galactose *in vitro* is inhibited by ethanol, but this inhibition is not reversed by menthol or progesterone. Depression by ethanol of galactose oxidation in human liver *in vitro* is reversed by menthol, as has been observed in other animal species.

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