# Antioxidant: A New Role for RU-486 and Related Compounds

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#### **Abstract**

RU-486 (17  $\beta$ -hydroxy 4-dimethylaminophenyl-17- $\alpha$ -propenyl estrone 4,9 diene-3-one; mifepristone) is suggested to act by binding to progesterone and glucocorticoid receptors. Based on its chemical nature, we anticipated that RU-486 may have potent antioxidant properties. We used the oxidation of LDL as our model system. RU-486 and a similar compound, onapristone, at 1-5-\(\mu\)M concentrations, decreased the formation of oxidized LDL. LDL isolated from plasma of subjects who were orally supplemented with RU-486 was resistant to oxidation, as compared to LDL isolated from control plasma.

The antioxidant effect of RU-486 appears to reside in the dimethylaminophenyl side chain moiety. Reduction of the A-ring of the steroid molecule had no effect on its antioxidant property. Analogs of RU-486 which lack the dimethylaminophenyl group, were without antioxidant activity. Levonorgestrel, which lacks the dimethylaminophenyl group failed to inhibit the oxidation of LDL even at  $100-\mu M$  levels. In contrast, ethinylestradiol and estradiol which do not possess the dimethylamino group, were able to inhibit the oxidation of LDL by virtue of their phenolic steroid "A" ring.

Thus RU-486, with its long half life, high plasma concentrations, association with lipoproteins, and ability to readily enter the cell may have additional intra- and extra-cellular antioxidant effects. (J. Clin. Invest. 1994. 94:1990-1995.) Key words: lipid peroxidation • low density lipoprotein • onapristone • antiprogestins • estradiol

## Introduction

RU-486 or mifepristone is a steroid derivative that possesses potent antiprogesterone and antiglucocorticoid activities (1, 2). In addition to its use in the termination of pregnancy (3, 4), recent studies have suggested that RU-486 is effective in providing relief from pain associated with endometriosis (5), and induced a marked reduction in leiomyoma volume (6). Further-

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more, tumors that have hormone receptors, such as breast cancer and meningiomas, have been reported to respond clinically to RU-486 (7-9). Its efficacy as an antiglucorticoid also has received attention in the treatment of patients with Cushing's syndrome (10). However, agonist-like proliferative effects have also been recognized in cultured breast cancer lines and in postmenopausal women (11, 12). Such effects have been attributed to the presence of increased levels of cAMP (13). Despite considerable progress in the understanding of the biochemistry and molecular biology of steroid hormone receptors and the interaction of antiprogestins with the receptor (1, 2, 14-17), very little is known about the potential nonhormonal actions of these compounds. It is efficiently absorbed from the intestine and plasma concentrations reach micromolar amounts (1, 2, 18, 19). We anticipated that RU-486 may interact with an oxidation system and may exhibit potent antioxidant properties, based on its chemical nature, its similarity to the antioxidant, Lazaroid U-74,500A (a lipophilic dimethyl- or ethyl- amino group), and the formation of mono- and di-demethylated and hydroxylated metabolic products (20, 21).

A plethora of evidence has suggested that oxidized LDL may play an important role in the development of arteriosclerotic lesion (22-33). This oxidation can be prevented either by the enrichment of the lipoprotein or of the cells with antioxidants (22, 23, 27).

In this study, we demonstrate that RU-486 and similar steroid analogs such as onapristone, that have a dimethylaminophenyl group, have potent antioxidant properties. We also provide evidence to suggest that the antioxidant activity resides in the nonsteroid moiety, namely the dimethylaminophenyl group of this molecule. We show that LDL isolated from patients who were given oral doses of RU-486 is resistant to oxidation when exposed to in vitro oxidation conditions.

### Methods

RU-486 was a gift from Roussel-Uclaf (Romainville, France). Onapristone and ZK 112,993 were generously provided by Schering, AG (Berlin, Germany). Ethinylestradiol and other chemicals were purchased from Sigma Chemical Co. (St. Louis, MO). The chemical structures of these compounds are given in Fig. 1. LDL was isolated from normal human donors and radioiodinated as described before (22, 23, 27). The LDL was subjected to oxidation in the presence of 5  $\mu$ M copper in phosphate-buffered saline in spectrophotometric cuvettes and the increase in absorption at 234 nm was followed continuously up to 6 h (34). Oxidation incubations with copper or with endothelial cells (rabbit aortic cell line, B4) were followed by established procedures by incubating 100  $\mu$ g LDL protein/ml in 2 ml Ham's F-10 medium (22, 23, 27, 34). RU-486 and other drugs were added in 0.1% of ethyl alcohol. Aliquots were taken for thiobarbic acid products (TBARS)<sup>1</sup> analysis and for macrophage degradation (27, 34). Mouse peritoneal macrophages were harvested by peritoneal lavage, and degradation incubations were carried out as described elsewhere using cells that were cultured overnight (27, 34).

<sup>1.</sup> Abbreviations used in this paper: EC, endothelial cell; ETYA, eicosatetraynoic acid; LOOH, lipid hydroperoxide; TBARS, thiobarbituric acid substances.

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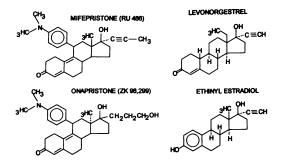


Figure 1. Structures of RU-486 and related compounds.

#### Results

In order to test the effect of RU-486 on the oxidation of LDL. the lipoprotein was subjected to oxidation with 5  $\mu$ M copper and the formation of conjugated diene was measured by following increase in absorption at 234 nm (34). As seen in Figs. 2 and 3, the control incubation generated an oxidized LDL that reached maximum increase in absorption at about 200 min. The lag phase, which represents the time at which antioxidants are consumed, was ~ 130 min. In the presence of increasing amounts of RU-486 (Fig. 2) or onapristone (Fig. 3), there was an increased lag time, and it was apparent that even as little as 0.5  $\mu$ M RU-486 was able to delay the oxidation rate. The maximum increase observed, at the end of the incubation, was not altered showing that the inhibitors did not affect the net availability of fatty acid molecules for oxidation in the system. The antioxidant effect was as potent as that observed with probucol (27, 35) or other phenolic oxidants, including estradiol, (36, 37) even though neither RU-486 nor onapristone have a phenolic function.

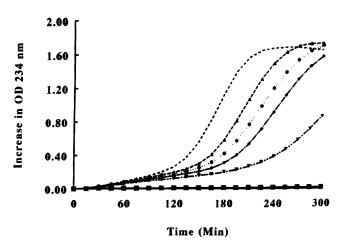


Figure 2. Effect of RU-486 on the oxidation of LDL by copper. LDL was incubated with 5  $\mu$ M copper in spectrophotometric cuvettes in the presence of specified concentrations of RU-486. OD at 234 nm was monitored continuously for 300 min. The figure represents results from a typical experiment from over six individual experiments. ( $\blacksquare$  - -, control LDL without copper; - - -, LDL incubated with 5  $\mu$ M copper;  $\blacktriangle$  - -, LDL incubated with 5  $\mu$ M copper and 0.1  $\mu$ M RU-486;  $\blacklozenge$  - -, LDL incubated with 5  $\mu$ M copper and 1.0  $\mu$ M RU-486;  $\blacktriangledown$  - -, LDL incubated with 5  $\mu$ M copper and 1.0  $\mu$ M RU-486;  $\blacktriangledown$  - -, LDL incubated with 5  $\mu$ M copper and 1.0  $\mu$ M RU-486;  $\blacktriangledown$  - -, LDL incubated with 5  $\mu$ M copper and 2.5  $\mu$ M RU-486.)

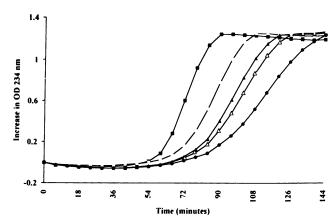


Figure 3. Effect of onapristone on the oxidation of LDL by copper. LDL was incubated with 5  $\mu$ M copper in spectrophotometric cuvettes in the presence of specified concentrations of onapristone. OD at 234 nm was monitored continuously for 300 min. The figure represents results from a typical experiment from over six individual experiments. ( $\blacksquare$ --, LDL incubated with 5  $\mu$ M copper; ---, LDL incubated with 5  $\mu$ M copper and 1  $\mu$ M onapristone;  $\blacktriangle$ --, LDL incubated with 5  $\mu$ M copper and 2  $\mu$ M onapristone;  $\vartriangle$ --, incubated with 5  $\mu$ M copper and 3  $\mu$ M onapristone;  $\bullet$ --, LDL incubated with 5  $\mu$ M copper and 4  $\mu$ M onapristone.)

The efficiency of RU-486 in inhibiting the oxidative modification of the LDL was studied using <sup>125</sup>I-labeled lipoprotein. LDL incubated with endothelial cells showed more than 2.5-fold increase in the electrophoretic mobility as compared to the control LDL. RU-486 decreased the formation of a more electronegative LDL. Similarly, the increase in mobility of LDL seen when LDL was oxidized with copper was also decreased by RU-486 (Table I). The formation of TBARS is a convenient

Table I. Effect of RU-486 on the Oxidative Modification of LDL by Endothelial Cells and Copper

Sample	Relative mobility (CM)	Macrophage degradation μg/5 h per mg cell protein	
		Medium	Cell associated
Native LDL	0.9	0.4	0.5
LDL incubated with EC	2.5	6.2	9.6
EC-LDL prepared in			
the presence of			
$2 \mu M$ RU-486	2.2	4.0	6.0
5 μM RU-486	1.7	1.9	2.9
Cu-LDL	2.4	6.9	10.3
Cu-LDL prepared in the presence of			
2 μM RU-486	1.7	1.9	2.8
5 μM RU-486	1.3	1.3	2.7

<sup>125</sup>I-LDL (100  $\mu$ g/ml) was incubated with confluent endothelial cells or with 5  $\mu$ M copper in Ham's F-10 medium in 60-mm dishes for 24 h at 37°C. The medium was harvested and tested for TBARS, electrophoretic mobility, and macrophage degradation. The values represent the average of a duplicate set of determinations from a representative experiment. Four independent experiments were performed in which the degree of inhibition varied ±14%.

Table II. Effect of RU-486 and Reduced RU-486 on the Oxidative Modification of LDL by 5  $\mu$ M Copper

Sample	Macrophage uptake $\mu g/5$ h per mg cell protein		
	Degradation	Cell associated	
Native LDL	1.3	1.1	
Cu-LDL	5.4	4.1	
Cu-LDL prepared in the			
presence of			
2 μM RU-486	1.2	1.4	
5 μM RU-486	0.9	1.3	
Cu-LDL prepared in the presence of			
2 μM Reduced RU-486	0.8	1.0	
5 $\mu$ M Reduced RU-486	0.6	1.2	

<sup>125</sup>I-LDL (100  $\mu$ g/ml) was incubated with confluent endothelial cells or with 5  $\mu$ M copper in Ham's F-10 medium in 60-mm dishes for 24 h at 37°C. The medium was harvested and tested for TBARS and macrophage degradation. The values represent average of a duplicate set of determinations from a representative experiment. Two separate experiments were performed.

means of measuring lipid peroxidation. The LDL incubated with 5  $\mu$ M copper generated over 55 nmol of TBARS and was efficiently degraded by the macrophages. Addition of 2–5  $\mu$ M RU-486 decreased the formation of TBARS and decreased the degradation by macrophages. This was not due to the effect of RU-486 on the intracellular metabolism of oxidized LDL in macrophages, as the cell-associated radioactivity or accumulation of oxidized LDL in macrophages was also decreased. The degradation of the LDL incubated in the presence of RU-486 during oxidation was similar to that of the control LDL, incubated in the absence of copper.

In order to test whether the antioxidant effect was due to the progesterone ring, RU-486 was subjected to reduction using sodium borohydride. The reduced RU-486 was still as effective as RU-486 in inhibiting the oxidation of LDL showing almost similar lag phases as compared to the unreduced RU-486 (data

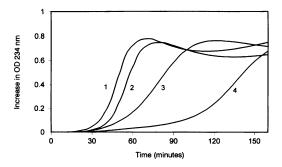


Figure 4. Effect of ZK 112993 on the oxidation of LDL. LDL was incubated with 5  $\mu$ M copper in spectrophotometric cuvettes in a total volume of 1 ml of PBS in the presence or absence of ZK 112993 or RU-486. The formation of conjugated dienes was followed by measuring the increase in absorption at 234 nm. The figure represents results from a typical experiment from three individual experiments. 1, control incubation; 2, incubation in the presence of 10  $\mu$ M ZK 112993; 3 and 4, incubation in the presence of 0.5 and 1  $\mu$ M RU-486.

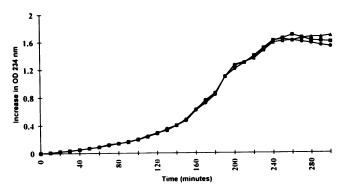


Figure 5. Effect of levonorgestrel on the oxidation of LDL by copper. LDL was incubated with 5  $\mu$ M copper in spectrophotometric cuvettes in the presence of specified concentrations of levonorgestrel. OD at 234 nm was monitored continuously for 300 min. The figure represents results from a typical experiment from three individual experiments. ( $\blacktriangle$ --, control incubation in the absence of levonorgestrel;  $\blacksquare$ --, incubation in the presence of 5  $\mu$ M levonorgestrel;  $\bullet$ --, incubation in the presence of 10  $\mu$ M levonorgestrel.)

not given). As shown in Table II, the inclusion of reduced RU-486 also prevented the formation of TBARS and macrophage degradation, suggesting again that the progesterone ring does not contribute to the antioxidant effect of RU-486. In fact, the reduced RU-486 was even more active as an antioxidant than RU-486 itself. Similar effects were noted when reduced RU-486 was incubated in a cell oxidation system utilizing mouse macrophages (results not shown).

RU-486 also has a side chain that contains a propenyl group. This propenyl group is an acetylenic bond similar to that found in the lipoxygenase inhibitor and antioxidant, namely eicosatetraynoic acid (ETYA). In order to distinguish whether the propenyl group contributes to the antioxidant effect or the antioxidant effect is due to the dimethyl aminophenyl group, we utilized two other compounds, levonorgestrel and ZK 112,993. These compounds are similar to RU-486, except that the dimethylaminophenyl group is replaced by a hydrogen atom and a acetophenyl side chain, respectively. These compounds did not show any significant antioxidant effects even at 10- $\mu$ M levels (Figs. 4 and 5). The addition of 10  $\mu$ M ZK 112,993 increased the lag

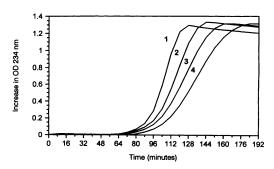


Figure 6. Effect on N,N'-dimethylaniline on the oxidation of LDL. LDL was incubated with 5  $\mu$ M copper in spectrophotometric cuvettes in the presence of specified concentrations of dimethylanilinelevonorgestrel. OD at 234 nm was monitored continuously for 300 min. The figure represents results from a typical experiment from a duplicate set of experiments. Numbers represent incubations that contained 0, 2, 4, and 6  $\mu$ M N,N'-dimethylaniline.

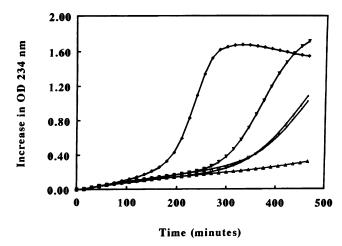


Figure 7. Effects of RU-486 and Ethinylestradiol on the oxidation of LDL by copper. LDL was incubated with 5  $\mu$ M copper in spectrophotometric cuvettes in the presence of specified concentrations of RU-486 or ethinylestradiol. OD at 234 nm was monitored continuously for 500 min. The figure represents results from a typical experiment from four individual experiments. ( $\bullet$ --, control incubation; 0--, incubated with 2.5  $\mu$ M RU-486;  $\bullet$ --, incubated with 5  $\mu$ M RU-486;  $\bullet$ --, incubated with 5  $\mu$ M ethinylestradiol.)

phase by less than 5 min. The addition of 1  $\mu$ M RU-486, on the other hand, tripled the lag phase. In contrast, clomiphene and N,N'-dimethylaniline, nonsteroid compounds with diethyland dimethylamino groups, respectively, were effective in decreasing the oxidation of LDL (Fig. 6). However, dimethylaniline was not as potent as RU-486 as an antioxidant, which may suggest additional factors such as solubility in a lipid environment. The lack of effect of levonorgestrel or ZK 112,993 was not due to a counteracting effect of the propenyl group in these compounds. As a control, we used ethinyl estradiol, which also possesses the propenyl group in addition to a phenolic hydroxyl group, and found it was capable of inhibiting the oxidation of the LDL (Fig. 7). Taken together, these results suggest that neither the progesterone ring nor the acetylenic triple bond themselves contributed to the antioxidant property. However, the dimethylaminophenyl group of RU-486 may provide an important antioxidant center for the molecule. The phenolic compound ethinyl estradiol inhibited the oxidation of LDL as expected.

Single dose studies have suggested that in human serum RU-486 is bound to  $\alpha$ -1-acid glycoprotein and the plasma concentration may be limited by the saturation of this protein (2, 18). On the other hand, onapristone does not appear to bind to this protein (1). However, its lipophilicity, may suggest that it may also be carried in lipoprotein fractions in the plasma, particularly upon long term administration. In order to test whether RU-486 is indeed carried in the lipoproteins, we isolated the LDL from two subjects who consumed 25 mg of RU-486 for 8 wk (studies of RU-486 for the treatment of endometriosis and leiomyomata were approved by the internal review board of the University of California, San Diego). LDL was also isolated from control subjects. The plasma from the RU-486 treated subjects had 7.3 and 11.7 nm of RU-486/ml as determined by spectroscopic methods. When subjected to oxidation, the LDL isolated from the RU-486 subjects was very resis-

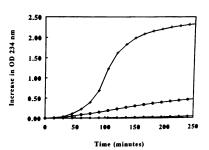


Figure 8. Oxidation of LDL from RU-486 subjects. LDL was isolated from two subjects who consumed RU-486, and was subjected to oxidation as described in the text. Two independent determinations were performed. x--, LDL from two control subjects; ◆ and ◆--, LDL from RU-486 subjects.

tant to oxidation in the presence of copper and showed lag phases approximately three to four times higher than the control (Fig. 8).

The inhibition of LDL oxidation by antioxidant enriched cells has been reported using an analog of probucol (27). RU-486 readily enters the cell and its site of antihormonal action is intracellular (1, 3, 11, 12). We, therefore, incubated RAW macrophage cells with RU-486 for 8 h and tested the ability of these cells to oxidize LDL after washing. As seen in Table III, cells that were incubated with LDL in Ham's F-10 medium in the presence of RU-486 oxidized the lipoprotein poorly. Since the cells were pretreated with RU-486 and no free RU-486 remained in the medium during the incubation with LDL, this may suggest that RU-486 may afford additional protection against oxidative damage to cells.

## **Discussion**

These studies indicate that RU-486 is clearly an antioxidant and may also have other effects, unrelated to its antiprogesterone or antiglucocorticoid effects. Its ability to inhibit the oxidation of LDL by copper in an in vitro system may suggest that its antioxidant activity may also be manifested in plasma compartment, unrelated to its antiprogesterone and antiglucocorticoid effects. Similar compounds, Lazaroids, particularly U-74,500A (38–40) have been found to be effective in controlling oxidation in cells and tissues that are independent of sterol regulation. However, U-74,500A, by virtue of its steroid nucleus is effective against proliferation of breast cancer cells, suggesting that the antioxidant property of RU-486 or onapristone may play an important role in its efficacy against breast cancer.

Detailed analysis of the molecular mechanisms by which

Table III. Oxidation of LDL by Cells Pretreated with RU-486

Incubation conditions	TBARS (% control)	
Untreated cells		
Cells treated with vehicle alone	98±3	
Cells pretreated with 25 $\mu$ M RU-486	59±11	
Cell pretreated with 50 µM RU-486	23±9	

Confluent RAW cells were washed and preincubated with RU-486 for 8 h at 37°C. The cells were washed and subsequently incubated, in the absence of RU-486, with 100  $\mu$ g/ml LDL for an additional 24 h in Ham's F-10 medium. The medium was analyzed for TBARS. The values represent mean  $\pm$  SD (n=4).

RU-486 affects progesterone receptor-mediated transcriptional regulation has failed to show any major differences between progesterone and its antagonists (1, 2, 14-17). The steroidmediated activation of gene transcription appears to involve a dual component transcription function of the receptors, a hormone-dependent and a hormone-independent (41-43) component. It is possible that the antioxidant component of RU-486 may also affect the hormone-independent interactions. Recently it was suggested that the presence of cAMP adversely affects the antagonist action of RU-486 and similar antiprogestins (13). Upon increasing the cellular levels of cAMP, antagonist-coupled progesterone receptors became strong transcriptional activators suggesting cAMP-responsive factors. Agents that elevate cAMP also activate nuclear factors, particularly NFkB (44, 45) and antioxidants protect cells against oxidative stress and prevent or regulate gene expression, particularly those that have regulatory domains available to nuclear factor NFkB (46-48). Thus the possibility that free or bound, intracellular RU-486, would suppress mechanisms by which intracellular cAMP and NFkB are elevated thereby further decreasing transcription appears promising.

The interaction of RU-486, an amino compound, with oxygen radicals is not surprising. Compounds such as diphenylphenylenediamine (DPPD) (31), the lazaroids (38-40), have been in use as antioxidants to prevent the oxidation of LDL. Demethylation of amines is a characteristic oxidation reaction (49, 50) and in vivo, such reactions characterize detoxification reactions. Major metabolism of RU-486 occurs via demethylation and hydroxylation reactions (20, 21). Indeed, a cytochrome P-450 isozyme system has been described that catalyzes these reactions, suggesting that RU-486 may indeed be capable of interacting with an oxidation system (51). In conclusion, the finding that RU-486 and related compounds are potent antioxidants may suggest additional, potential therapeutic applications, as these compounds have already been tested for their safety and effectiveness. In fact, tomoxifen, another antiestrogen with a diethylaminogroup was recently shown to inhibit the oxidation of LDL at  $\mu$ M concentrations (52). More over glucocorticoids affect the expression of the acetyl LDL receptor, a receptor that recognizes among other ligands, oxidized LDL and bacterial lipopolysaccharides (53, 54). RU-486 with its antioxidant and antiglucocorticoid action will significantly alter the course of oxidation and macrophage function. It is difficult to assess how important this property is in mediating the anti-hormonal action; however, their ready entry into cellular compartments provides a significant advantage over existing lipophilic antioxidants.

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