

# Renal Bicarbonate Wasting during Phosphate Depletion

## A POSSIBLE CAUSE OF ALTERED ACID-BASE HOMEOSTASIS IN HYPERPARATHYROIDISM

LAWRENCE W. GOLD, SHAUL G. MASSRY, ALLEN I. ARIEFF, and JACK W. COBURN

*From the Medical Research Institute and Renal and Hypertension Service of the Cedars-Sinai Medical Center, and the Departments of Medicine of Cedars-Sinai Medical Center, Veterans Administration Wadsworth Hospital Center, and University of California Los Angeles School of Medicine, Los Angeles, California 90048*

**ABSTRACT** With hyperparathyroidism, serum bicarbonate ( $\text{HCO}_3^-$ ) is low, urinary excretion of  $\text{HCO}_3^-$  is increased and the apparent  $T_m$  for  $\text{HCO}_3^-$  is reduced. These findings have been ascribed to a direct renal action of parathyroid hormone (PTH). Since hypophosphatemia and phosphate depletion may occur in hyperparathyroidism, it is possible that phosphate depletion could account for the abnormal renal  $\text{HCO}_3^-$  handling. To test this possibility, renal reabsorption of  $\text{HCO}_3^-$  was evaluated in dogs before and after phosphate depletion. Serum  $\text{HCO}_3^-$  was significantly lower in phosphate depleted dogs than in normal animals, and serum  $\text{HCO}_3^-$  was directly related to serum phosphorus. Both the threshold at which  $\text{HCO}_3^-$  appeared in the urine and the  $T_m$  for  $\text{HCO}_3^-$  were reduced during phosphate depletion. Intracellular pH of muscle was significantly higher in phosphate depleted dogs than in normals and the pH returned to normal after phosphate repletion. These data show that phosphate depleted dogs, which probably have physiological hypoparathyroidism, display abnormalities in both serum  $\text{HCO}_3^-$  and its renal handling which are similar to those seen in hyperparathyroidism, supporting the concept that the PTH-induced alterations in  $\text{HCO}_3^-$  homeostasis may be due to phosphate depletion. The latter could alter cell metabolism, resulting in reduced intracellular  $\text{H}^+$  concentration, which may then impair  $\text{H}^+$  secretion by the renal tubules and decrease their ability to reabsorb  $\text{HCO}_3^-$ . Consequently,  $T_m$   $\text{HCO}_3^-$  and serum  $\text{HCO}_3^-$  fall.

Dr. Shaul G. Massry is an Established Investigator of the American Heart Association.

Dr. Allen I. Arieff is a Clinical Investigator of the Veterans Administration.

Received for publication 5 March 1973 and in revised form 4 June 1973.

## INTRODUCTION

Clinical and experimental observations indicate that there is a relation between the activity of the parathyroid glands, on the one hand, and acid-base homeostasis and renal handling of bicarbonate on the other. Thus, hyperchloremic acidosis has been noted occasionally in patients with primary hyperparathyroidism (1-3), and others have found impaired urinary acidification in response to an acid load in such patients (4-6). In addition, mild systemic acidosis, low levels of serum bicarbonate ( $\text{HCO}_3^-$ ), excessive urinary loss of  $\text{HCO}_3^-$ , and impaired urinary acidification have been reported in patients with osteomalacia and secondary hyperparathyroidism (6-9). These abnormalities were usually reversed following treatment of the hyperparathyroidism (10). Furthermore, the levels of serum  $\text{HCO}_3^-$  are usually elevated both in patients with hypercalcemia which is not due to hyperactivity of the parathyroid glands and in patients with hypoparathyroidism (1, 11-14). Finally, Barzel (13, 14) found that patients with hypoparathyroidism had mild metabolic alkalosis. These observations led Muldowney et al. (10) to postulate that parathyroid hormone directly decreases the renal reabsorption of bicarbonate resulting in metabolic acidosis. Support for such a postulate is found in experimental observations that the acute administration of parathyroid extract (PTE)<sup>1</sup> produces a rise in urinary pH and an increase in urinary excretion of  $\text{HCO}_3^-$  (15-18). However, Kurtzman, Karlinsky, and Sager (19) found that the administration of PTE or dibutyl cyclic AMP, an analogue of the mediator of renal action of parathyroid hormone

<sup>1</sup> Abbreviations used in this paper: GFR, glomerular filtration rate; PTE, parathyroid extract; PTH, parathyroid hormone.

(PTH), caused only a transient and slight increase in  $\text{HCO}_3^-$  excretion and failed to reduce  $\text{HCO}_3^-$  reabsorption when evaluated during  $\text{HCO}_3^-$  loading.

It is possible that abnormalities of bicarbonate homeostasis in patients with hyperparathyroidism may be secondary to the metabolic consequences of the long-term action of excessive quantities of parathyroid hormone rather than being a direct result of the hormones action on the renal tubule. With sustained action of parathyroid hormone, phosphaturia and hypophosphatemia occur and phosphorus depletion may develop. The present study was undertaken to evaluate whether phosphate depletion, per se, can affect renal bicarbonate handling and acid-base homeostasis.

## METHODS

Experiments were carried out in 18 adult female mongrel dogs, weighing 13–22 kg. Phosphate depletion was produced in eight dogs by the feeding of a low phosphate diet and the administration of aluminum hydroxide gel (kindly supplied by Dr. Earl Lewis of the Wyeth Laboratories, Philadelphia, Pa.) as described previously from this laboratory (20). The diet was identical during control, depletion and repletion periods with regard to its content of sodium, potassium, calcium and magnesium; only the phosphate content was varied.

Experiments for measurement of maximum reabsorption rate for bicarbonate and determination of the renal threshold for bicarbonate were carried out in six animals during control and depletion periods as follows: Food was withheld for 16–18 h before each study but the animals were allowed free access to water. An oral water load, equal to

5% of body weight, was given by stomach tube at 8:00–8:30 a.m. on the day of the experiment. 45–60 min later, the dog was anesthetized with intravenous pentobarbital. A cuffed endotracheal tube was inserted, and respiration was maintained with a Harvard respirator (Harvard Apparatus Co., Millis, Mass.) adjusted to control  $\text{PCO}_2$ . The animal's rectal temperature was measured and maintained stable with blankets and electric heating pads. After the intravenous injection of 500 mg of creatinine in an 8% solution, a creatinine sustaining solution was given at 2.0 ml/min. Another solution of 2.5% dextrose in water was administered at a rate adjusted so that the total amount of fluid infused was equal to urine flow. Urine was collected anaerobically under mineral oil from an indwelling catheter in the bladder, and blood samples were obtained anaerobically in syringes containing heparin from a catheter in the femoral artery. Manual pressure over the suprapubic area was utilized to empty the bladder at the end of each period. After a 45 min interval for equilibration, three control clearance periods of 7–20 min duration were obtained. Over the next 200 min, the serum level of bicarbonate ( $\text{HCO}_3^-$ ) was elevated in a step-wise manner by increments of 2–5 meq/liter; this was carried out by the intravenous injection of 15–44 meq of  $\text{NaHCO}_3$  as a priming bolus in a solution containing 880 meq/liter; after each priming dose,  $\text{NaHCO}_3$  was added to the sustaining solution of creatinine in an amount to provide 60–240 meq/liter of  $\text{HCO}_3^-$ . This quantity of  $\text{NaHCO}_3$  was calculated to maintain the elevation of serum  $\text{HCO}_3^-$  produced by the preceding priming injection. After 30 min of equilibration at each successively higher level of serum  $\text{HCO}_3^-$ , clearance periods of 5–10 min duration were obtained. Serum bicarbonate eventually reached levels of 27–43 meq/liter. During each experiment, an attempt was made to prevent wide variation in  $\text{PCO}_2$  from the outset until the completion of

TABLE I  
Effect of Phosphate Depletion on Serum Bicarbonate and Renal Handling of Bicarbonate

Dog no.	Diet	Duration of diet	$\text{C}_{\text{Cr}}$	$\text{S}_{\text{P}}$	$\text{PCO}_2$		$\text{SHCO}_3^-$		$\frac{T_m \text{HCO}_3^-}{\text{C}_{\text{Cr}}} \times 100$	Total solids		$\text{S}_{\text{K}}$	
					C	$T_m$	C	$T_m$		C	$T_m$	C	$T_m$
		days	ml/min	mg/100 ml	mm Hg		meq/liter			%	%	meq/liter	
1	N	29	73.7	3.8	36	40	26.6	38.6	3.72	6.8	6.2	3.1	2.9
	L	95	65.2	1.3	36	41	23.0	36.3	2.91	7.0	6.9	3.2	3.0
2	N	27	50.0	2.9	39	39	27.1	34.3	3.32	7.0	6.7	3.4	2.5
	L	93	41.3	1.1	36	38	22.0	33.0	2.75	7.2	7.1	2.9	2.7
3	N	26	48.3	3.6	43	45	26.5	35.1	3.35	7.2	6.8	3.3	2.3
	L	66	39.8	0.4	36	46	22.4	42.6	3.04	6.9	6.5	3.1	2.5
4	N	25	84.9	4.4	38	48	23.3	40.0	3.49	6.8	6.7	2.8	3.1
	L	139	60.0	2.9	35	48	19.3	40.9	2.90	7.0	6.7	2.5	2.8
5	N	37	54.3	2.7	36	42	21.1	37.5	3.05	7.2	6.6	2.7	2.7
	L	40	57.4	0.6	33	37	20.1	35.1	2.29	7.2	6.9	2.9	2.9
6	N	21	91.1	3.0	37	38	22.5	33.2	2.70	7.7	7.8	2.7	2.7
	L	85	97.1	0.5	33	41	19.1	26.9	2.17	7.6	7.7	3.0	2.8
Mean	N		67.1	3.4	38	42	24.5	36.5	3.27	7.2	6.8	3.0	2.7
$\pm$ SE			$\pm 7.6$	$\pm 0.3$	$\pm 1$	$\pm 2$	$\pm 1.0$	$\pm 1.1$	$\pm 0.14$	$\pm 0.2$	$\pm 0.2$	$\pm 0.1$	$\pm 0.1$
	L		60.1	1.1	35	40	21.0	35.8	2.68	7.2	7.0	2.9	2.8
			$\pm 8.5$	$\pm 0.4$	$\pm 1$	$\pm 3$	$\pm 0.7$	$\pm 2.3$	$\pm 0.14$	$\pm 0.1$	$\pm 0.2$	$\pm 0.1$	$\pm 0.1$
P			NS	<0.001	0.05	NS	<0.02	NS	<0.02	NS	NS	NS	NS

Each data point represents the mean of three to five clearance periods. The values for  $\text{PCO}_2$  and serum bicarbonate ( $\text{SHCO}_3^-$ ) are indicated during the control part of each experiment (C) and later when  $T_m$  was being measured. Other abbreviations: N, normal phosphate diet; L, phosphate depletion diet;  $\text{C}_{\text{Cr}}$ , exogenous creatinine clearance;  $\text{S}_{\text{P}}$ , serum phosphorus;  $\text{SHCO}_3^-$ , serum bicarbonate;  $T_m \text{HCO}_3^-$ , maximum reabsorptive rate for bicarbonate;  $\text{S}_{\text{K}}$ , serum potassium.

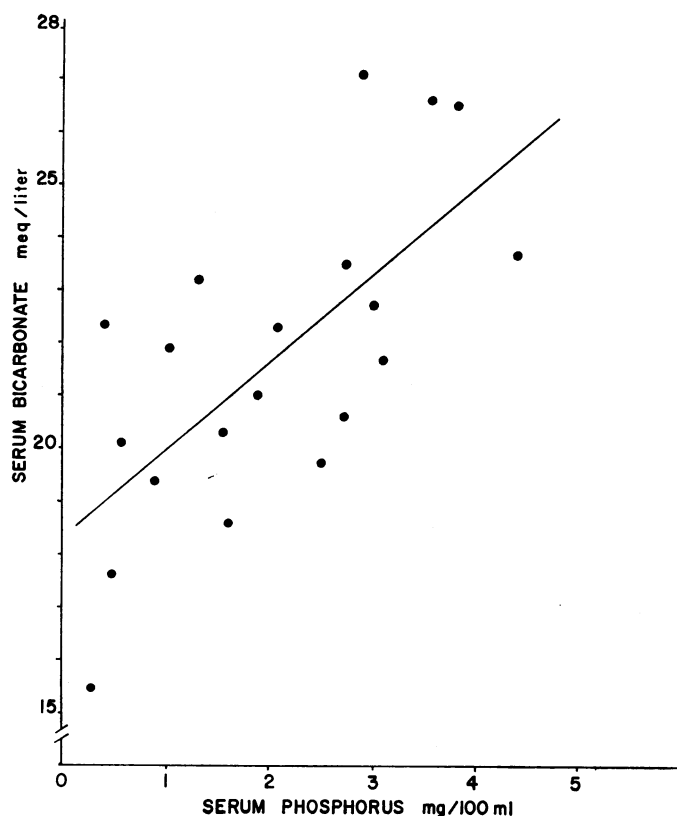


FIGURE 1 The relation between the levels of serum bicarbonate and phosphorus during the production of phosphate depletion.

the study; in most experiments, the changes were 8 mm Hg or less (Table I). During the latter part of the experiments when maximal rates of  $\text{HCO}_3^-$  reabsorption occurred, ventilation was closely regulated so that the  $\text{Pco}_2$  was similar in the same animal during both the control and phosphate depleted state; the variation was 5 mm Hg in one dog and 3 mm or less in the others (Table I). Also, the volume of fluid and quantity of sodium given to each animal was similar in experiments carried out during the control and phosphate depleted states.

Intracellular pH was measured in seven additional dogs receiving a normal diet of Purina dog chow, and in eight dogs during phosphate depletion (including all six described above); in four of the latter group, the measurements of intracellular pH of muscle were repeated 64–87 days after phosphate repletion. The method for measuring intracellular pH was similar to that reported by Waddell and Butler (21) and Adler, Roy, and Relman (22) utilizing  $[\text{C}]\text{DMO}$  (55-dimethyl-2,4-oxazolidinedione) and non-labeled DMO as a carrier. Extracellular space was measured by  $^{35}\text{SO}_4$ , and  $^{35}\text{SO}_4$  activity was separated from  $^{14}\text{C}$  activity by precipitation of  $^{35}\text{SO}_4$  with barium chloride as described previously from our laboratory (23). Both isotopes were obtained from the New England Nuclear Corp., Boston, Mass.

The analytical methods for creatinine, phosphate, sodium, and potassium are reported elsewhere (24). Total solids in plasma were measured by refractometry (Goldberg Refractometer, American Optical Corp., Buffalo, N. Y.). The pH and  $\text{Pco}_2$  in arterial blood and urine were measured utilizing

a blood microsystem acid-base analyzer (model BMS 3-PHM 71; Radiometer, Copenhagen). The concentrations of bicarbonate in plasma and urine were calculated from the Henderson-Hasselbach equation utilizing the following factors: Solubility coefficients for  $\text{CO}_2$  in plasma and urine of 0.0301 and 0.0309, respectively; a pK of 6.10 for plasma, and a pK for urine calculated from its ionic strength according to the formula,  $\text{pKa} = 6.33 - 0.5 \sqrt{\text{Na}^+ + \text{K}^+}$ , with the concentrations of Na and K given in equivalents per liter (25); corrections were not made for Donnan factors.

## RESULTS

The animals appeared healthy and maintained or even gained body weight during the period of phosphate depletion. The effect of phosphate depletion on serum  $\text{HCO}_3^-$  and the renal handling of this ion are shown in Table I and Figs. 1, 2, and 3. With the development of phosphate depletion, serum  $\text{HCO}_3^-$  levels fell in each animal, and the mean value of  $21.0 \pm 0.7$  (SE) meq/liter noted in the phosphate depleted animals was significantly lower than that observed before phosphate depletion ( $24.5 \pm 1.0$  meq/liter),  $P < 0.02$ . Furthermore, there was a direct correlation between the serum concentrations of  $\text{HCO}_3^-$  and phosphorus with  $y = 1.64x + 18.30$ ,  $r = 0.68$ ,  $P < 0.001$  (Fig. 1).

Fig. 2 depicts the quantity of  $\text{HCO}_3^-$  appearing in the urine with increasing concentrations of serum  $\text{HCO}_3^-$  in experiments before and after phosphate depletion. These data were obtained before and during the early periods of infusion of  $\text{NaHCO}_3^-$  prior to the achievement of the apparent maximal rate for  $\text{HCO}_3^-$  reabsorption. These thresholds for  $\text{HCO}_3^-$  were lower in each animal during phosphate depletion than in control studies. As filtered  $\text{HCO}_3^-$  was raised, its reabsorption increased gradually and then stabilized, reaching an apparent  $T_m$  level. This maximal rate of  $\text{HCO}_3^-$  reabsorption was lower in each dog during phosphate depletion than in control studies, with mean values of  $2.68 \pm 0.14$  and  $3.27 \pm 0.14$  meq/min per 100 ml GFR ( $P < 0.02$ ), respectively. The relation between  $\text{HCO}_3^-$  reabsorption and its filtered load for all clearance periods during all experiments with bicarbonate infusion is shown in Fig. 3. At comparable filtered loads of  $\text{HCO}_3^-$  (above 2.3 meq/min per 100 ml GFR), the reabsorption of  $\text{HCO}_3^-$  was lower during phosphate depletion than in control studies.

The intracellular pH in the phosphate depleted animals was higher than values obtained in seven normal

dogs (Table II, Fig. 4), with a mean value of intracellular pH of 7.02 ( $95 \pm 7$  nmol/liter  $\text{H}^+$ ) in the phosphate depleted animals compared with 6.86 ( $136 \pm 4$  nmol/liter  $\text{H}^+$ ) in normal dogs,  $P < 0.01$ . After phosphate repletion, intracellular pH returned to 6.84 ( $146 \pm 9$  nmol/liter  $\text{H}^+$ ), a mean value not different from normal but significantly different from the value during phosphate depletion,  $P < 0.01$ .

## DISCUSSION

The results of the present study demonstrate that phosphate depletion is associated with a fall in serum bicarbonate and a reduction in both the threshold at which bicarbonate appears in the urine and the apparent maximal tubular reabsorptive capacity for bicarbonate. These data indicate that phosphate depletion per se can cause abnormalities in bicarbonate homeostasis which are similar to those described in humans with primary (1-6) or secondary (6-9) hyperparathyroidism. There is indirect evidence to indicate that the activity of the parathyroid glands is inhibited during phosphate depletion. Thus, rats show hypoplasia of the parathyroid glands

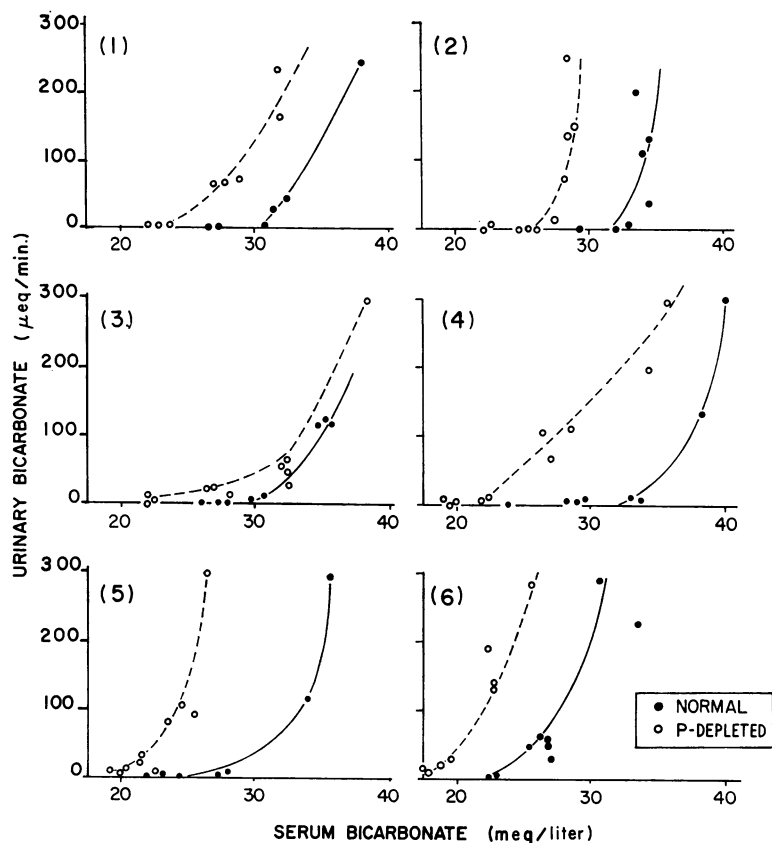


FIGURE 2 The relation between the amount of bicarbonate appearing in the urine as serum bicarbonate was elevated by bicarbonate infusion in experiments before (●) and after (○) phosphate depletion. The lines were drawn by visual approximation.

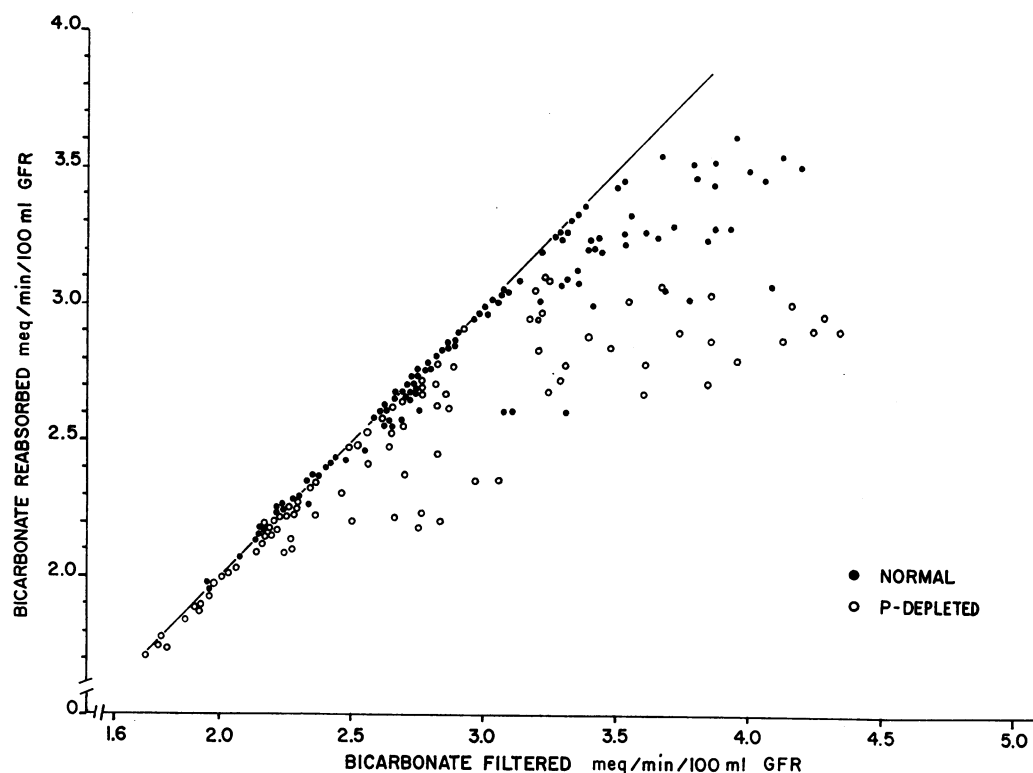


FIGURE 3 The relationship between the quantity of bicarbonate reabsorbed and that filtered for all clearance periods during all experiments with bicarbonate infusion both before (●) and after (○) phosphate depletion.

during phosphate depletion (26); also serum calcium levels remain normal or slightly elevated after removal of the parathyroid glands in phosphate depleted rats (26, 27) and dogs (20, 28), indicating that serum

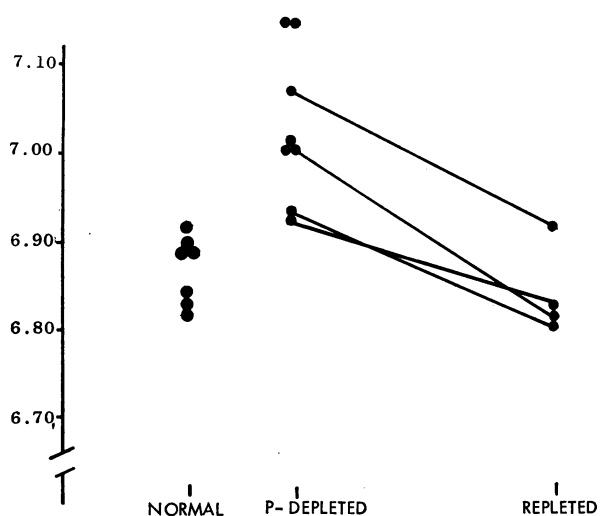


FIGURE 4 The effect of phosphate depletion and repletion on intracellular pH. The lines connect values obtained in the same animal.

calcium was maintained at normal levels in the absence of parathyroid hormone in the phosphate depleted state. Moreover, an increase in blood calcium to normal has been observed in humans with hypoparathyroidism during phosphate depletion (28). Such inhibition of the parathyroid glands during phosphate depletion is probably caused by transient or persistent elevation of serum calcium, a feature noted in rats and dogs during phosphate depletion (24, 25). It is, therefore, reasonable to conclude that the abnormalities of bicarbonate homeostasis observed in the present study, are associated with reduced rather than increased activity of the parathyroid glands despite the lack of measurement of parathyroid hormone in the blood of these animals; hence, the alterations in bicarbonate regulation are probably related to phosphate depletion per se.

It has been suggested that parathyroid hormone can cause an immediate and direct inhibition of the tubular reabsorption of bicarbonate. Nordin (17) gave PTE to humans and Helman, Au, and Bartter (18) to dogs and man. They found an immediate increase in urinary pH and an increase in  $\text{HCO}_3^-$  excretion after the intravenous injection of the extract. However, these changes could have resulted from alterations in filtered load of  $\text{HCO}_3^-$  secondary to an augmented glomerular filtration rate

TABLE II  
Effect of Phosphate Depletion on Intracellular pH

Dog no.	Diet	Duration of diet	S <sub>p</sub>	pHe	pHi	P <sub>PO<sub>2</sub></sub>
		days	mg. 100 ml			mm Hg
Normal dogs						
A	PC		5.1	7.37	6.84	34
B	PC		4.4	7.36	6.83	34
C	PC		4.3	7.37	6.89	35
D	PC		3.9	7.34	6.88	34
E	PC		3.3	7.32	6.82	40
F	PC		4.4	7.33	6.92	38
G	PC		2.7	7.40	6.88	35
Phosphate depleted dogs						
1	L	93	0.5	7.36	7.15	36
2	L	86	1.9	7.38	7.07	36
	N	64	3.4	7.43	6.92	31
3	L	78	0.9	7.41	7.00	35
	N	75	4.8	7.39	6.81	33
4	L	119	0.2	7.40	7.00	34
5	L	33	0.9	7.32	6.93	38
	N	75	5.2	7.33	6.82	37
6	L	65	0.5	7.39	6.93	30
	N	87	4.6	7.35	6.80	41
7	L	48	0.6	7.45	7.15	35
8	L	63	0.3	7.21	7.01	39

Abbreviations: PC, Purina Dog Chow (Ralston Purina Co., St. Louis, Mo.); L, phosphate depletion diet; N, normal phosphate (repletion) diet; S<sub>p</sub>, serum phosphorus level; pHe, blood pH; pHi, intracellular pH. Phosphate depleted dogs are designated by the same numbers as in Table I.

(GFR) or be due to spontaneous diurnal variations in HCO<sub>3</sub><sup>-</sup> excretion. Kleeman and Cooke (16) found an inconsistent effect of PTE on HCO<sub>3</sub><sup>-</sup> excretion, and Muldowney et al. (6, 10) studied the effect of PTE infusion in four subjects and reported data suggesting that an increase in HCO<sub>3</sub><sup>-</sup> excretion could occur independent of changes in GFR or diurnal variations in HCO<sub>3</sub><sup>-</sup> excretion. However, it may be more appropriate to study the effect of PTE during HCO<sub>3</sub><sup>-</sup> loading in order to critically evaluate a possible role of the hormone in reducing renal reabsorption of HCO<sub>3</sub><sup>-</sup>. Under such circumstances, Kurtzman, Karlinsky, and Sager (19) found no effect of either PTE or dibutyl cyclic AMP on tubular reabsorption of HCO<sub>3</sub><sup>-</sup>.

It is well established that parathyroid hormone causes phosphaturia due to decreased renal tubular reabsorption of phosphate, leading to a reduction in serum phosphorus (28, 29), and phosphate depletion may ensue. These observations and the results of the present study suggest that the alterations in HCO<sub>3</sub><sup>-</sup> homeostasis noted in patients with hyperparathyroidism may be largely mediated

by phosphate depletion rather than by a direct action of parathyroid hormone on the renal handling of HCO<sub>3</sub><sup>-</sup>.

The precise mechanisms responsible for altered HCO<sub>3</sub><sup>-</sup> reabsorption in phosphate depletion have not been delineated. The reduced tubular reabsorption of HCO<sub>3</sub><sup>-</sup> cannot be accounted for by factors known to affect its renal handling such as changes in serum potassium, alterations in Pco<sub>2</sub> (30), or variations in the degree of expansion of extracellular fluid volume (31, 32) since these parameters were either similar or carefully controlled during studies carried out in the control and phosphate depleted states. The present observation of an elevated intracellular pH during phosphate depletion may provide an explanation for the reduced tubular reabsorption of HCO<sub>3</sub><sup>-</sup>. Since tubular reabsorption of HCO<sub>3</sub><sup>-</sup> is dependent on secretion of hydrogen ion (H<sup>+</sup>) by the tubular cell (30), a decrease in intracellular concentration of H<sup>+</sup> would be expected to impair its secretion and, hence, reduce HCO<sub>3</sub><sup>-</sup> reabsorption. The mechanisms responsible for the increase in intracellular pH during phosphate depletion remain unknown.

The results of the present study may permit construction of the following sequence of events whereby phosphate depletion affects HCO<sub>3</sub><sup>-</sup> homeostasis: Through some as yet undefined mechanism, phosphate depletion causes a decrease in intracellular H<sup>+</sup> concentration. This intracellular alkalosis may impair tubular H<sup>+</sup> secretion and HCO<sub>3</sub><sup>-</sup> reabsorption. Increased urinary losses of HCO<sub>3</sub><sup>-</sup> lead to a fall in serum HCO<sub>3</sub><sup>-</sup> and mild acidosis.

Phosphate depletion has been reported to cause alterations in the renal tubular transport of other actively transported substances: Thus, there is decreased tubular reabsorption of calcium and magnesium (20), and, under certain circumstances, of sodium as well (33). Moreover, the maximal reabsorptive capacity for glucose is also reduced during phosphate depletion (unpublished observations). It is possible that the reduced reabsorption of HCO<sub>3</sub><sup>-</sup> represents another manifestation of altered renal tubular transport mechanisms caused by phosphate depletion.

#### ACKNOWLEDGMENTS

We are thankful for Mrs. Eleanor Chapman, Miriam Bick, and Anne Moran for their technical help and for Mrs. Catherine Hoyt for her secretarial assistance.

This work was supported by U. S. Public Health Service grants RR05468, AM05630, and AM5383.

#### REFERENCES

1. Wills, M. R., and G. K. McGowan. 1964. Plasma-chloride levels in hyperparathyroidism and other hypercalcemic states. *Br. Med. J.* 1: 1153.
2. Lafferty, F. W. 1956. Pseudohyperparathyroidism. *Medicine (Baltimore)*. 45: 247.

3. Dolman, M., and U. Barzel. 1971. Primary hyperparathyroidism. Diagnosis and Treatment. *N. Y. State J. Med.* 71: 1201.
4. Wrong, O., and H. E. F. Davies. 1959. The excretion of acid in renal disease. *Q. J. Med.* 28: 259.
5. Fourman, P., D. McComkey, and J. W. G. Smith. 1960. Defects of water reabsorption and of hydrogen-ion excretion by the renal tubules in hyperparathyroidism. *Lancet.* 1: 619.
6. Muldowney, F. P., R. Freaney, and T. McGeeney. 1968. Renal tubular acidosis and amino-aciduria in osteomalacia of dietary or intestinal origin. *Q. J. Med.* 37: 517.
7. York, S. E., and E. R. Yendt. 1966. Osteomalacia associated with renal bicarbonate loss. *Can. Med. Assoc. J.* 94: 1329.
8. Muldowney, F. P., R. Freaney, and P. Brennan. 1965. A case of osteomalacia in renal tubular acidosis associated with occult idiopathic steatorrhea; the effect of vitamin D on renal tubular hydron transport. *Ir. J. Med. Sci.* 6: 435.
9. Muldowney, F. P., J. F. Donohoe, R. Freaney, C. Kampff, and M. Swan. 1970. Parathormone-induced renal bicarbonate wastage in intestinal malabsorption and in chronic renal failure. *Ir. J. Med. Sci.* 3: 221.
10. Muldowney, F. P., D. V. Carroll, J. F. Donohoe, and R. Freaney. 1971. Correction of renal bicarbonate wastage by parathyroidectomy. *Q. J. Med.* 40: 487.
11. Heinemann, H. O. 1965. Metabolic alkalosis in patients with hypercalcemia. *Metab. (Clin. Exp.).* 14: 1137.
12. Thomas, W. C., Jr., T. B. Connor, and H. G. Morgan. 1958. Some observations on patients with hypercalcemia exemplifying problems in differential diagnosis, especially in hyperparathyroidism. *J. Lab. Clin. Med.* 52: 11.
13. Barzel, U. S. 1969. Systemic alkalosis in hypoparathyroidism. *J. Clin. Endocrinol. Metab.* 29: 917.
14. Barzel, U. S. 1971. Parathyroid hormone, blood phosphorus, and acidbase metabolism. *Lancet.* 1: 1329.
15. Ellsworth, R., and W. M. Nicholson. 1955. Further observations upon changes in the electrolytes of urine following the injection of parathyroid extract. *J. Clin. Invest.* 14: 823.
16. Kleeman, C. R., and R. R. E. Cooke. 1951. The acute effects of parathyroid hormone on the metabolism of endogenous phosphate. *J. Lab. Clin. Med.* 38: 112.
17. Nordin, B. E. C. 1960. The effect of intravenous parathyroid extract on urinary pH bicarbonate and electrolyte excretion. *Clin. Sci.* 19: 311.
18. Hellman, D. E., W. Y. W. Au, and F. C. Bartter. 1965. Evidence for a direct effect of parathyroid hormone on urinary acidification. *Am. J. Physiol.* 209: 643.
19. Kurtzman, N. A., M. L. Karlinsky, and D. S. Sager. 1973. Effects of infusion of cyclic AMP and parathyroid hormone on renal bicarbonate reabsorption. *Clin. Res.* 21: 283.
20. Coburn, J. W., and S. G. Massry. 1970. Changes in serum and urinary calcium during phosphate depletion: Studies on mechanisms. *J. Clin. Invest.* 49: 1073.
21. Waddell, W. J., and T. C. Butler. 1959. Calculation of intracellular pH from the distribution of 5,5-dimethyl-2,4-oxazolidinedione (DMO). Application to skeletal muscle of the dog. *J. Clin. Invest.* 38: 720.
22. Adler, S., A. Roy, and A. S. Relman. 1965. Intracellular acid-base regulation. I. The response of muscle cells to changes in CO<sub>2</sub> tension or extracellular bicarbonate concentration. *J. Clin. Invest.* 44: 8.
23. Bradbury, M. W. B., M. Villamil, and C. R. Kleeman. 1968. Extracellular fluid, ionic distribution and exchange in isolated frog brain. *Am. J. Physiol.* 214: 643.
24. Massry, S. G., J. W. Coburn, and C. R. Kleeman. 1969. The influence of extracellular volume expansion on renal phosphate reabsorption in the dog. *J. Clin. Invest.* 48: 1237.
25. Hastings, A. B., and J. Sendroy, Jr. 1925. The effect of variation in ionic strength on the apparent first and second dissociation constants of carbonic acid. *J. Biol. Chem.* 65: 445.
26. Stoerk, H. C., and W. H. Carnes. 1945. The relation of the dietary Ca:P ratio to serum calcium and to parathyroid volume. *J. Nutr.* 29: 43.
27. Shikita, M., S. Tsurufuji, and Y. Ito. 1962. Adaptation in renal phosphorus excretion under the influence of parathyroids; a study of ureterally catheterized rats. *Endocrinol. Jap.* 9: 171.
28. Freeman, S., and F. C. McLean. 1941. Experimental rickets. Blood and tissue changes in puppies receiving a diet very low in phosphorus, with and without vitamin D. *Arch. Pathol.* 32: 387.
29. Lotz, M., E. Zisman, and F. C. Bartter. 1968. Evidence for a phosphorus-depletion syndrome in man. *N. Engl. J. Med.* 278: 409.
30. Pitts, R. F. 1968. Physiology of the kidney and body fluids. Yearbook Medical Publishers, Inc., Chicago, Ill. 2nd edition. 185.
31. Purkerson, M. L., H. Lubowitz, R. W. White, and N. S. Bricker. 1969. On the influence of extracellular fluid volume expansion on bicarbonate reabsorption in the rat. *J. Clin. Invest.* 48: 1754.
32. Kurtzman, N. A. 1970. Regulation of renal bicarbonate reabsorption by extracellular volume. *J. Clin. Invest.* 49: 586.
33. Gold, L. W., S. G. Massry, C. M. Rothman, and J. W. Coburn. 1973. Effect of phosphate depletion on renal handling of calcium and sodium. *Clin. Res.* 21: 282.