

Recoupling the Na-K Pump

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ABSTRACT Human red blood cells display under appropriate circumstances a ouabain-sensitive K-K exchange when the flux measurements are made using radioisotopes. Such an exchange complicates measurements of the coupling of Na outflux to K influx in cells which are partially depleted of energy sources by deprivation of glucose since the K-K exchange has been found to be increased in depleted cells. When the measurements of flux are made by estimating net cation movements chemically, it is found that glucose deprivation results in a fall in *both* ouabain-sensitive Na outflux and ouabain-sensitive K influx. Since both fluxes fell in concert, there is no reason for believing that the fluxes are not coupled or that the source of ATP for the Na outflux is different from that for the K influx.

INTRODUCTION

Since the first demonstration that the Na outflux from human red blood cells is reduced when extracellular K is removed (1), a considerable amount of evidence has accumulated to support the concept that the active Na outflux and active K influx are coupled in the sense that movement of Na out of the cell through the pump must be accompanied by a simultaneous movement of K into the cell. It has been demonstrated that the relation between the magnitude of the active K influx and the extracellular K concentration is the same as the relation between the active Na outflux and the extracellular K

concentration (2), and that both the active Na outflux and K influx increase as the intracellular Na concentration is increased (3). Careful measurements have shown that, for every three Na ions transported out of the cell, two K ions are transported into the cell, and one ATP molecule is split (3-5).

In a recent report, Feig, Segel, Shohet, and Nathan (6) state that, when human red cells are starved for short periods of time, the ouabain-sensitive Na outflux falls but the ouabain-sensitive K influx is little affected. They point out that, in the starved red cells, glyceraldehyde-3-phosphate is depleted and hence little ATP is formed by the phosphoglycerate kinase (PGK)¹ reaction. They suggest that ATP formed by this reaction is preferentially used as the source of energy for the Na outflux. They further suggest that, since 2,3-diphosphoglycerate (2,3-DPG) remains high for a time, there is sufficient substrate for the pyruvate kinase reaction and ATP from this source serves as the source of energy for the active K influx. If it is true that the energy source for the active Na outflux and active K influx is not the same, then the two fluxes must be independent and the evidence that they are coupled must be fortuitous.

It is possible, however, that a different explanation might be found for the observations of Feig et al. (6). In their studies, the measurements of the Na and K fluxes were made using the radioactive isotopes ²⁴Na and ⁴²K. Under these circumstances, the ouabain-sensitive outflux of ²⁴Na and influx of ⁴²K will include not only the active outflux of Na and influx of K, but also the ouabain-sen-

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¹ *Abbreviations used in this paper:* 2,3-DPG, 2,3-diphosphoglycerate; FDP, fructose 1,6-diphosphate; GAPDH, glyceraldehyde-P-dehydrogenase; K_i, intracellular K; K_o, K in extracellular fluid; Na_i, intracellular Na; PCMBs, parachloromercuribenzenesulfonic acid; PGK, phosphoglycerate kinase; triose-P, triose phosphate.

TABLE I
Effect of Intracellular K Concentration (K_e) on the Ouabain-Sensitive K Influx in Cells with Very Low Na Concentrations (Na_e)

Na_e	K_e	Ouabain-sensitive K influx (mean \pm SE; $n = 4$)	Ouabain-sensitive Na outflux (mean \pm SE; $n = 4$)
mmoles/liter erythrocytes		mmoles/liter erythrocytes, 1 hr	
1.2	89.1	0.445 \pm 0.015	0.022 \pm 0.003
1.2	0.9	0.026 \pm 0.003	0.091 \pm 0.004

Na_e and K_e were altered by exposure to PCMBs solutions containing glucose 5 mM, and dithiothreitol solutions containing (mM): glucose 5, adenine 3, and inosine 2. The bulk of the cation in the low K_e cells was made up of choline. The measurements of K influx were made in glycylglycine buffered $MgCl_2$ -sucrose solutions with K_0 0.67 mM, and the Na outflux measurements in the same solution except that K_0 was 4.0 mM.

sitive Na-Na and K-K exchange (7-9). If, in the starved cells, the magnitude of the K-K exchange increases as the magnitude of the active K influx decreases, one might expect that the total ouabain-sensitive ^{42}K influx might remain unchanged. In the starved cells used by Feig et al. (6), it would be expected that ATP levels fall and inorganic phosphate levels rise as phosphorylated intermediates are broken down; these have been found to be conditions which increase the magnitude of the K-K exchange (7-9).

The present paper reports the results of experiments designed to determine whether the active Na outflux and active K influx decrease to the same extent in starved cells in which K-K exchange cannot occur.

METHODS

The solutions used, manipulation of the cells, procedures used for the measurement of unidirectional K influx and Na outflux, and the methods used for the measurement of intracellular Na and K concentrations have been described (10). Purified choline chloride was obtained from Hoffman-Taff, Springfield, Mo. and was not further processed.

Alteration of intracellular cation concentrations was accomplished by a modification of a method described by Garrahan and Rega (11). Washed cells were suspended at about 5% hematocrit in a solution containing (mM): PO_4 3.4, Mg^{++} 1.0, parachloromercuribenzenesulfonic acid (PCMBs) 0.1, Cl^- 147, glucose 10, and varying concentrations of Na, K, and choline to make up a total of 150 mM. Sucrose was added at concentrations varying from 0 (all choline solutions) to 55.6 mM (choline-free solutions) to maintain constant cell volumes. In some cases adenine and/or inosine were included. The suspensions were incubated at 4°C for 36 hr, and the solutions were changed every 12 hr. At 36 hr, the cells were separated from the solutions and resuspended in a solution identical to the PCMBs solution except that PCMBs was omitted and dithiothreitol 2 mM was

included. The cells were incubated in this solution for 1 hr at 37°C, separated from the solution, washed three times in $MgCl_2$ solution (107 mM), and used as appropriate.

Glycolytic intermediates were estimated by enzymatic reactions which result in the stoichiometric oxidation of NADH; NADH concentration was measured fluorometrically (12). Neutralized 12% perchloric acid extracts of red cell suspensions were used for the determinations. ATP was determined with glyceraldehyde-P-dehydrogenase (GAPDH) and PGK; 2,3-DPG with GAPDH, PGK and P-glycerate mutase; and fructose 1,6-diphosphate + triose phosphate (FDP + triose-P, reported as equivalent concentration of triose-P) with GAPDH, triose-P isomerase and aldolase.

RESULTS AND DISCUSSION

The existence in human red blood cells of a ouabain-sensitive K outflux which, in Na-free solutions, depends on the presence of K in the extracellular solution (K_e) is clear evidence for a ouabain-sensitive K-K exchange in these cells (8, 9). If there is a ouabain-sensitive K-K exchange, it should be possible to demonstrate a ouabain-sensitive K influx which is independent of intracellular Na (Na_e), but does depend on intracellular K (K_e). Post and Sen (7) found a ouabain-sensitive K influx in cells with Na_e reduced so low as to eliminate any ouabain-sensitive Na outflux; this influx did not occur in inorganic phosphate-free cells. To demonstrate that this K influx in Na free cells is a K-K exchange, it is necessary to show that it depends on K_e . An experiment designed to show such a dependence is summarized in Table I. Cells were prepared with very low Na_e ; half had high concentration K_e , and half were almost K-free. The ouabain-sensitive K influx was much higher in the cells containing K than in the K-free cells. The K influx could not have been in exchange for a Na outflux since the ouabain-sensitive Na outflux in these cells was very small. The simplest explanation for a ouabain-sensitive K outflux dependent on K_e and a ouabain-sensitive K influx dependent on K_e is the mechanism of K-K exchange diffusion (13).

Since K-K exchange diffusion is a phenomenon which is seen only with the use of radioisotopes (13), it cannot complicate the measurement of net cation fluxes. Therefore net changes of Na and K were measured in cells which were altered to contain very low concentrations of K_e (to increase the reliability of the determinations) and slightly high concentrations of Na_e ; Na_e and K_e in the solutions in which the measurements were made were chosen so that the cation gradients were small and the ouabain-insensitive fluxes minimal. One group of cells (repleted cells) had fairly normal concentrations of FDP + triose P, and the other group had low concentrations; the ATP and 2,3-DPG content of both groups was about normal. During the measurement of cation fluxes, the depleted cells were incubated in solutions containing glucose, and the depleted cells in glucose-free solutions.

TABLE II
Ouabain-Sensitive Na Loss and K Gain in Low K_e Cells

	Experiment I			Experiment II		
	Repleted	Depleted	Δ (Repleted – depleted)	Repleted	Depleted	Δ (Repleted – depleted)
Na _e , mmoles/liter erythrocytes	25.2	24.9		21.9	25.7	
K _e , mmoles/liter erythrocytes	5.9	5.7		2.4	3.0	
ATP, mmoles/liter erythrocytes	1.42	1.31		1.06	0.57	
2,3-DPG, mmoles/liter erythrocytes	5.29	4.29		4.42	4.12	
FDP + Triose-P (as triose-P); mmoles/liter erythrocytes	0.46	0.003		0.50	0.007	
Na _o , mM	25.1	25.1		23.5	23.5	
K _o , mM	4.2	4.2		4.3	4.3	
Ouabain-sensitive Na loss, mmoles/liter erythrocytes, 2 hr	–7.4	–3.6	–3.8	–9.7	–3.2	–6.5
	±0.58	±0.27		±0.22	±0.40	
Ouabain-sensitive K gain, mmoles/liter erythrocytes, 2 hr	+5.1	+1.9	+3.2	+7.3	+1.7	
	±0.40	±0.09		0.19	0.15	+5.6
Ratio (Na loss/K gain)	1.45	1.89	1.19	1.33	1.88	1.16

Red cell cation contents were altered by incubation in PCMBS solutions. In experiment I, the PCMBS solution contained (mM): glucose 10, adenine 2, and inosine 0.5; in experiment II the PCMBS solution contained (mM): glucose 10 and adenine 2. In each case, the repleted cells were incubated in dithiothreitol solutions containing (mM): glucose 10 and inosine 1; the depleted cells were incubated in dithiothreitol solutions free of substrate. The values of Na_e, K_e, ATP, 2,3-DPG, and FDP + Triose P were those at the start of the experiment. The bulk of the intracellular cation was choline. The solutions in which the repleted cells were incubated for measurement of cation fluxes contained glucose 10 mM, and the depleted cells were incubated in glucose-free solutions. The bulk of the electrolyte in the solution was choline chloride. Samples for determination of cation content were taken 15 min and 2 hr and 15 min after the cells were added to the solution; incubation was carried out at 37°C. The values for the cation fluxes are means ±SE, *n* = 6.

Table II presents the ouabain-sensitive net changes in Na_e and K_e that occur over a 2 hr period in two of five experiments. In the depleted cells the ouabain-sensitive Na outflux and K influx are both lower than in the repleted cells, and the ratio of the amount by which the Na outflux fell as a result of depletion to the amount by which the K influx fell (1.19 and 1.16 in experiments I and II, respectively) was a little greater than one. Thus when K-K exchange is eliminated, the response of the ouabain-sensitive Na outflux and of the ouabain-sensitive K influx to the removal of glucose is the same.

Stoichiometric coupling of the Na-K pump at a ratio of 3 Na out to 2 K in requires that the ratio of ouabain-sensitive Na outflux to ouabain sensitive K influx be 1.5. The mean ratio for repleted cells from five experiments similar to those reported in Table II was 1.45±0.033 (SE), which is not different from 1.5. However, the ratio for depleted cells in the same experiments was 1.83±0.16 (SE) but the variability was too great to permit evaluation of the significance of the deviation from 1.5. When the ratio of the fall in Na outflux due to depletion to the fall in K influx was calculated, the ratio was 1.31±0.06 (SE). It is uncertain that these deviations are significant; if they are, either of two mechanisms might account for them. First, it may be that, in depleted cells, a part of the Na outflux becomes uncoupled from the K

influx. In this event, it should be possible to demonstrate a ouabain-sensitive Na outflux into K-free solutions using depleted cells; such a flux could not, in fact, be demonstrated in either repleted or depleted cells. A second possible explanation is that the Na/K coupling ratio is in some way dependent on the ATP concentration with more Na ions exchanged for each K ion in depleted cells; this possibility was not further evaluated.

On the basis of studies of the mechanism by which ouabain inhibits glycolysis in hemolysates containing membranes, Parker and Hoffman (14) suggested that ADP produced by the Na,K-ATPase may be compartmentalized within the membrane, and that this ADP is the substrate for membrane-bound PGK. Proverbio and Hoffman (15) have recently presented evidence which they interpret as indicating that the source of ATP for the Na,K-ATPase reaction comes preferentially from a membrane compartment. Feig et al. (6) proposed that ATP derived from the PGK reaction may be preferentially used for the active Na outflux since the ouabain-sensitive Na outflux fell at a time when triose phosphate, the source of substrate for the PGK reaction, was low but the total intracellular ATP + 2,3-DPG was still high. Since part of the red cell PGK is membrane bound (16), this would fit with the findings of Parker and Hoffman (14) and Proverbio and Hoffman (15). Since

it is clear that only a fraction of the ATP consumed by the red cells is used for cation transport, there is no way of telling from the present experiments whether or not ATP from the PGK reaction is preferentially used by the pump. There is no basis, however, for believing that the source of ATP used for Na outflux is different from the source of that used for K influx, especially since the Na-K pump is driven by intracellular ATP in reconstituted ghosts independent of the source reaction for ATP (17).

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