ETHANOL METABOLISM AND PRODUCTION OF FREE ACETATE IN THE HUMAN LIVER

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Lundsgaard solved the problem of identifying the organ in which the first stage of the metabolism of ethanol takes place (1). He clearly demonstrated that in cats the liver accounts for 90 per cent or more of the ethanol removed from the circulating blood. He also observed that an organic acid is produced in the perfused liver during alcohol metabolism.

The question of the pathway of ethanol metabolism, however, is still partly unsolved. Several possibilities may be considered. There is general agreement that oxidation to acetaldehyde is the first stage in the metabolic conversion of ethanol, but at least three possibilities may be envisaged for the fate of acetaldehyde. Its oxidation may lead to the formation of acetyl coenzyme A, (acetyl-CoA, a common intermediate in the breakdown of carbohydrate, amino acids, and fats) or to free acetate, or it could possibly enter into a number of condensation reactions (2). If acetyl-CoA were the primary product of acetaldehyde oxidation, acetoacetate (and β -hydroxybutyrate) might be formed, in addition to oxidation to carbon dioxide and water and incorporation into other substances.

Experiments in which an enzymic method was used for acetate determination have shown that the concentration of acetate in peripheral blood of human subjects increases to about 20 times the normal level when ethanol is present in the organism, whereas neither fasting nor the intake of a meal rich in fats influences the blood acetate concentration (3).

On the other hand, Forsander and Räihä (4), while confirming the presence of acetate in the blood of rats given ethanol, found a considerable increase in acetoacetate. Lundsgaard (1) found no significant increase in ketone bodies in perfused cat liver.

The present experiments were designed to ascertain to what extent free acetate is formed in the liver from ethanol in normal human subjects and to study other metabolic changes during ethanol combustion, including the influence of ethanol on the splanchnic oxygen consumption. Seligson and co-workers (5), in an abstract, have reported a somewhat similar study comprising two normal and two cirrhotic patients.

METHODS

Nine males and one female with no known hepatic or metabolic disorder were examined (Table I). They were fasted for 12 hours and premedicated with 200 mg of phenobarbitone. A Cournand catheter was placed in one of the right hepatic veins and a polythene catheter in the femoral artery. Blood was drawn simultaneously from the two catheters into heparinized syringes.

The experiments comprised a control and an ethanol period, each lasting about 30 minutes. Twenty minutes before the control period a priming dose of 150 mg of sulfobromophthalein (BSP) was given, and an infusion of BSP was started (at an average rate of 4.7 mg per minute). During the control period 4 to 6 samples were taken from both catheters for the determination of BSP and 2 to 3 for the determination of acetate; in some cases 2 to 3 samples were taken for glucose, ketones, and lactate; in the middle of the period, O2 saturation, O2 tension, pH, and CO2 tension were determined in one sample. The ethanol period was started with a single injection of 70 mmoles of ethanol, except in Subject 2 in whom the injection caused venospasm. Simultaneously, a constant infusion of 2.7 mmoles per minute of ethanol was started and continued for the rest of the experiment. During this period BSP and ethanol were determined in 5 to 8 samples; acetate and, in some cases, ketones, glucose, and lactate were determined in 4 to 6 samples; and O₂ saturation, O₂ tension, pH, and CO₂ tension in one sample in the middle of the period.

BSP (6), ethanol (7), acetate (8), acetoacetate, β -hydroxybutyrate (9), glucose (10), and lactate (11) were determined in heparin plasma by the methods indicated. From the hematocrit values the results were calculated

TABLE	I
Experimental	subjects

No.	Age	Height	Weight	Sex
	yrs	cm	kg	
1	35	172	79	M
2	37	174	78	M
3	22	172	62	M
4	27	169	64	M
5	45	159	40	F
6	56	161	38	\mathbf{M}
7	21	177	68	M
8	32	170	63	M
9	19	172	61	M
10	18	180	63	M

for whole blood. Hepatic blood flow was calculated as indicated by Bradley, Ingelfinger, Bradley and Curry (12). Separate experiments were made to measure the distribution of ethanol and acetate between erythrocytes and plasma. Ethanol was present in proportion to the amount of water, whereas acetate was distributed in accordance with the Donnan equilibrium, since the concentration in the water phase of the erythrocytes was approximately 90 per cent of that in the water phase of the surrounding plasma.

Oxygen saturation was determined by a reflectometer; O_2 tension by a falling mercury drop electrode (13); pH and pCO₂ by Astrup's method (14); and hemoglobin concentration by a spectrophotometer. The oxygen content of whole blood was calculated from the oxygen saturation and the oxygen capacity of the sample.

RESULTS

Figure 1 shows a typical experiment. It was found in all cases that the arterial blood acetate concentration reached a nearly constant level, about 0.8 mM, soon after the ethanol infusion was started. The arteriohepatic venous acetate and ethanol concentration differences showed only small variations within each experiment. In one experiment (no. 2) the arterial ethanol concentration rose steadily from 2 to 4 mM during the experiment. The arteriohepatic venous differences for ethanol and acetate were also constant in this case.

Table II gives a summary of 10 experiments. In all cases the output of acetate from the splanchnic area was more than 50 per cent of the ethanol disappearance. The average fall in hepatic blood flow in the ethanol period is statistically significant (p < 0.05). The small drop in the average splanchnic oxygen consumption on the other hand is not statistically significant (p > 0.1). The splanchnic oxygen consumption and ethanol re-

moval appeared to be correlated (r=+0.69), and the same was the case with acetate production and hepatic blood flow (r=+0.73). These correlation coefficients are statistically significant at the 5 per cent but not at the 1 per cent level. The number of experiments is too limited to warrant an interpretation of these findings. The arteriohepatic venous difference in total ketone body concentration was found to be insignificant in all experiments except one (no. 5), where the net output of acetoacetate and β -hydroxybutyrate was 0.10 mmole in the control period and 0.18 mmole in the ethanol period.

The ratio of acetoacetate to β -hydroxybutyrate in the blood from the hepatic vein decreased significantly during the ethanol period in all experiments, as seen in Table III. The lactate concentration in arterial blood increased during the ethanol period in the three cases examined. This increase is not caused exclusively by production of lactate in the liver, as seen in a typical experiment (Table IV). In the two other experiments a small output (about 0.14 mmole per minute) from the splanchnic area was measured, while the arterial lactate concentration rose at a rate of about 0.02 mM per minute.

The blood glucose concentrations in the hepatic venous blood showed considerable fluctuation during the experiments, rendering an assessment of

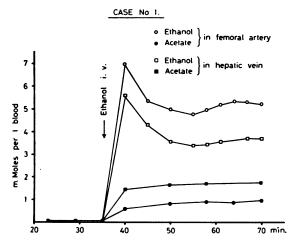


FIG. 1. TYPICAL EXPERIMENT; PLASMA CONCENTRA-TIONS OF ETHANOL AND ACETATE IN THE FEMORAL AR-TERY AND HEPATIC VEINS. At the time indicated by the arrow intravenous administration of ethanol was started with a single injection, followed by a constant infusion (experiment 1).

TABLE 11

Determination of ethanol consumption and acetate production in the splanchnic area of the 10 subjects *

		Splanch.		Av. art. plasma	Av. art. plasma		calculated, le blood	Ethanol consump., splanch.		Acetate produc
Exp. no.	Hepatic Exp. no. blood flow (A-V O ₂ diff.	A-V splanch. O ₂ diff. area		acetate ethanol conc.		Ethanol Acetate		splanch. area	Ethanol consump ×100
	L/min†	mmoles / L†	mmoles / min†	mМ	mM	mM	mM	mmoles / min	mmoles / min	%
1	2.14	2.06	4.37							
	1.74	2.59	4.50	0.81	4.70	1.23	-0.70	2.14	1.21	56
2	1.36	2.54	3.48							
	1.23	2.77	3.40	0.78	2.59	1.46	-0.80	1.80	0.99	55
3	(1.95)‡	2.06	(4.02) [‡]							
	1.95	2.23	4.33	0.78	1.70	0.91	-0.76	1.78	1.47	83
4	1.47	1.56	2.28							
	1.30	1.65	2.14	0.59	2.78	0.89	-0.93	1.15	1.20	104
5	1.53	2.64	4.02							
	1.47	2.46	3.61	0.95	7.52	1.56	-0.88	2.29	1.29	56
6	1.61	2.14	3.44							
	1.62	2.46	3.98	0.80	3.28	0.82	-0.84	1.33	1.35	101
7	1.17	2.46	2.86							
	1.31	1.92	2.50	0.84	4.67	0.67	-0.61	0.87	0.80	92
8	1.77	1.47	2.59							
	1.48	1.43	2.10	0.43	3.08	0.84	-0.76	1.24	1.13	91
9	1.46	2.23	3.26							
	1.14	2.32	2.63	0.82	3.11	1.39	-0.73	1.58	0.83	53
10	2.23	1.96	4.42							
	1.47	2.10	3.08	0.79	2.97	0.89	-0.54	1.31	0.79	60
Average	1.64	2.10	3.48							
Ö	1.42	2.19	3.22	0.76	3.64	1.07	-0.76	1.55	1.11	75

^{*} The concentrations given are average values for several samples (see Methods).

† The upper figures refer to the control periods.

net splanchnic glucose balance unreliable (15). In spite of this, the splanchnic glucose output in most cases tended to decrease during the ethanol period, as illustrated in Table V. In most cases the pH of the hepatic venous blood was reduced during the ethanol infusion (on an average from 7.40 to 7.35), showing that acid was formed in the splanchnic area. From the pH and pCO₂

measurements, the total carbon dioxide content of the blood was estimated by means of the nomogram of Singer and Hastings (16). The results indicate that the respiratory quotient of the splanchnic area was reduced during the ethanol period, but the method is not sufficiently accurate to permit conclusions regarding the absolute figures.

TABLE III

Changes in the ratio acetoacetate/β-hydroxybutyrate in blood plasma during ethanol metabolism *

	Control period		Ethanol period		period as $\%$ of control	
Exp. no.	Arterial	Hep. vein	Arterial	Hep. vein	Arterial	Hep. vein
4	0.12	0.24	0.16	0.14	133	58
5	0.35	0.42	0.29	0.25	83	60
6	0.26	0.43	0.21	0.19	81	44
7	0.22	0.30	0.16	0.15	74	50
8	0.22	0.18	0.25	0.17	114	94
9	1.0	1.1	1.1	0.61	110	55
10	0.23	0.32	0.21	0.17	91	53
				Average	98	59

^{*} In most experiments 2 samples in the control period and 4 to 6 samples in the ethanol period were used. The A-V difference for total ketone bodies generally did not change significantly when the ethanol infusion was started.

[‡] No determination of the blood flow was made in this control period. The figure for the experimental period has been applied in the calculation of the oxygen consumption.

TABLE IV

Changes in lactate concentration in arterial and hepatic venous blood plasma during ethanol metabolism; experiment 10

Time	Lact concent				
from start of exp.	Arterial	Hep. vein	Diff.	Average	
min	m l	M	m	ıM	
Control peri	od				
25	1.13	1.37	-0.24		
35	1.28	1.27	+0.01	-0.04	
45	1.29	1.18	+0.11		
thanol per	iod				
55	1.62	1.67	-0.05		
65	1.83	1.82	+0.01	0.04	
75	2.04	2.16	-0.12	-0.06	
85	2.16	2.25	-0.09		

INTERPRETATION AND DISCUSSION

The previously observed increase in blood acetate (3) during ethanol metabolism does not show to what extent the ethanol that is metabolized is liberated into the blood stream as free acetate. In the present experiments, however, it is clear that a major part of the ethanol is oxidized to free ace-Compared with most other studies on ethanol metabolism the blood ethanol concentrations employed in the present study are low. This was done partly to permit an accurate determination of the arteriohepatic venous difference and partly because it was considered desirable to study the ethanol metabolism in organisms showing no signs of intoxication. Another metabolic effect of ethanol, the inhibition of the hepatic elimination of galactose, in previous experiments was found to be fully developed at similar concentrations of ethanol (17), and it has been observed (3) that the acetate concentration in venous blood is independent of the blood ethanol concentration within wide limits—another indication in support of the widely accepted view that ethanol metabolism in man does not depend upon the concentration in the range investigated.

The question now arises: What happens to the part of the ethanol that is not recovered as acetate in the venous blood from the liver? The most obvious explanation, perhaps, is that it is oxidized to CO_2 and water in the liver; but this assumption meets with considerable opposition. Forsander, Räihä and Suomalainen (18) examined the fate of C^{14} -labeled ethanol in perfused rat liver

and found that the CO₂ formed contained no appreciable radioactivity. Also, experiments by Bartlett and Barnet (19) with liver slices gave evidence that only a small part of the ethanol is converted to CO₂ in the liver. These findings are in agreement with experiments by Lundquist and Svendsen (20) with rat liver homogenates, in which nearly all the ethanol metabolized was recovered as free acetate.

On the other hand, there seems to be little doubt that the liver can activate (and consequently oxidize) acetate, but at a much lower rate than the oxidation of ethanol to acetate. In peripheral tissues, however, acetate is readily metabolized, as is also evident in the present experiments.

For the interpretation of experiments in which hepatic venous blood is examined it must be kept in mind that a large part of the blood (the portal blood) has passed through other organs before reaching the liver. It seems reasonable to suppose that while the liver produces, almost exclusively, acetate from the available ethanol without further oxidation, the organs of the portal area may oxidize some of the acetate present in the arterial blood feeding these organs. In this case the measured acetate formation will be less than the ethanol consumption in the splanchnic area.

In Table VI the oxygen consumption calculated for these processes is compared with the measured oxygen consumption in the splanchnic area. On the average, 74 per cent of the oxygen consumption has been used in the oxidation of ethanol and acetate. It must be noted that the

TABLE V

Changes in glucose concentration in arterial and hepatic
venous blood plasma during ethanol metabolism;
experiment 10

Time	Glu- concen	cose tration			
from start of exp.	Arte- rial	Hep. vein	Diff.	Average	
min	m	M	m	:M	
Control perio	d				
25	4.75	5.88	-1.13		
35	4.76	5.00	-0.24	-0.51	
45	4.87	5.04	-0.17		
thanol perio	od				
55	5.20	5.04	+0.16		
65	5.27	4.89	+0.38		
75	5.19	5.09	+0.10	+0.14	
85	4.68	4.92	-0.24		

89

101

74

Splanchnic oxygen consumption in relation to the metabolism of ethanol *							
	Oxygen consumptio	Total calc.					
Exp. no.	Of ethanol to acetate (liver)	Of acetate to CO ₂ + H ₂ O (extrahep. spl. area)	O_2 consump. as % of measured O_2 consump.				
	mmoles/min %†	mmoles/min %†	%				

1.86

0.62 0 1.52 0 0.14 0.22

1.04

0.85

TABLE VI

Splanchnic oxygen consumption in relation to the metabolism of ethanol*

† Per cent of the measured oxygen consumption in the splanchnic area.

49

1.80

1.55

In this experiment the net increase in formation of ketone bodies of 0.26 mmole/min was taken into account.

calculations of Table VI imply the assumption that about one-third of the acetate elimination in the body takes place in the extrahepatic splanchnic organs. The figures for the extrahepatic mesenteric oxygen uptake calculated to account for the acetate disappearance do not seem unreasonably high in any of the experiments. Tygstrup, Winkler, Mellemgaard and Andreassen (21) found that, on the average, 34 per cent of the splanchnic oxygen was consumed outside of the liver. The possibility that acetoacetate and \(\beta\)-hydroxybutyrate are formed in the liver from ethanol has been ruled out. An increase in ketone body formation during ethanol infusion was observed in only one experiment, and this was quantitatively unimportant.

10

Average

A third acetate pathway which should be considered is the incorporation into other substances, notably lipids. Such processes undoubtedly do take place, and equally well from ethanol and acetate (22, 23), but to a very minor degree, which can at most explain a small fraction of the acetate deficit in some of the experiments (24), if results obtained in animals are accepted as valid in man. It is therefore considered most probable that all of the ethanol is oxidized to free acetate in the liver, and that the reason for the low recovery of acetate in the hepatic venous blood in some experiments is

mainly oxidation in the extrahepatic splanchnic

25

Lactate analyses were made in order to decide whether some of the ethanol oxidation takes place as a dismutation reaction—i.e., coupled to the reduction of pyruvate to lactate. Such a process would mean that the liver could deal with a larger quantity of ethanol without complete oxidation of the hydrogen. The increase in blood lactate, however, was not caused exclusively by the liver but may indicate that the peripheral tissues prefer acetate rather than lactate as a fuel.

The finding that the ratio of acetoacetate to β -hydroxybutyrate decreases during ethanol metabolism to about half the control value in hepatic venous blood is another indication that the ratio DPNH/DPN is increased in the liver during alcohol metabolism (25, 26). A similar observation was made by Büttner, Portwich and Engelhardt (27), who found that the ratio of pyruvate to lactate, which is also a DPN-dependent redox system, diminished in rats given ethanol. these findings strongly suggest that the change in the state of oxidation of the coenzymes is a very important consequence of ethanol metabolism. It is indeed possible that this phenomenon is responsible for the radical changes in metabolism of the liver which must take place as soon as ethanol

^{*} The figures are calculated on the assumption that all the ethanol not accounted for as acetate is completely oxidized to CO₂ and water. Since the acetate oxidation may have taken place largely in the extrahepatic part of the splanchnic area, this has been listed separately.

[§] In this experiment the figures given for the acetate formation are minimal values. The actual values may have been considerably higher, the oxygen requirement for complete oxidation, therefore, correspondingly lower.

is present in the circulating blood. An example of this change is the decline of galactose utilization (17).

The magnitude of the ethanol consumption in the splanchnic area in these experiments may seem rather low, being 4.3 g per hour as compared with a generally accepted figure for the ethanol elimination of about 7 g per hour. The results are, however, in agreement with those of Larsen (28). In both cases the experimental subjects were fasted for at least 12 hours before the experiment. Fasting has been shown in animals (29, 30) to reduce the rate of ethanol combustion. It is not probable that the low rate is a consequence of the low concentration employed, since similar low combustion rates were recorded in experiments at higher ethanol concentration (20 mM) (28). The observation that oxygen consumption does not change significantly when ethanol infusion is started, taken along with the finding that the ATP concentration of liver is not diminished by ethanol administration (27), sheds some light on the question of oxidative phosphorylation from extramitochondrial DPNH. The alcohol dehydrogenase responsible for the oxidation of ethanol is known to be exclusively present in the cytoplasm (31). It is therefore suggested that the yield of high energy bonds from DPNH is the same whether the DPNH is formed inside the mitochondria or not.

From Table VI it is calculated that, if 35 per cent of the total oxygen consumption in the splanchnic area takes place outside the liver (21), about 75 per cent of the oxygen consumption of the liver is used for ethanol metabolism. If the assumption made in Table VI that acetate is oxidized exclusively in the extrahepatic part of the splanchnic area is not justified, the figure will be even higher (up to 100 per cent).

This raises the question of what mechanism causes the "normal" metabolic processes to be suppressed, and also what consequences this change may have for the normal functions of the liver. As to the first question it is perhaps natural to regard the increased DPNH/DPN ratio as a possible clue, but this has not as yet been established.

Like Castenfors, Hultman and Josephson (32), we could not confirm Mendeloff's (33) observation that the hepatic blood flow was increased during ethanol metabolism. Castenfors and colleagues

found a small reduction in flow, which they thought might be unrelated to the ethanol given. In our cases the fall in flow was statistically significant, while in seven control experiments, in which the hepatic blood flow was observed for a similar interval, there was no significant reduction in the period corresponding to the ethanol period of the present experiments. Since other circulatory parameters (e.g., the cardiac output) have not been examined, the cause of the reduction in the hepatic blood flow cannot be stated.

It must therefore be concluded that no reaction, either metabolic or vasomotor, takes place to compensate the large amount of oxygen required by the liver for the oxidation of ethanol.

SUMMARY

- 1. The disappearance of ethanol and the formation of acetate in the splanchnic area were measured in human subjects through analysis of arterial and hepatic venous blood. At the same time liver blood flow and splanchnic oxygen consumption were determined. A quantity of acetate corresponding to 50 to 100 per cent of the ethanol metabolized was found in blood from the hepatic veins. The arterial acetate concentration reached a constant level of about 0.8 mM (50 μ g per ml) shortly after the administration of ethanol.
- 2. The splanchnic oxygen consumption was not significantly changed by the infusion of ethanol, whereas the hepatic blood flow showed a small but significant decrease.
- 3. Formation of ketone bodies by the liver was not significantly changed when ethanol was given, but the ratio acetoacetate/ β -hydroxybuty-rate decreased to about half the control value.
- 4. Lactate concentration in the arterial blood increased during ethanol infusion, but this was not exclusively caused by an increased output from the liver.
- 5. The significance of the findings in relation to the regulation of liver metabolism is discussed.

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