

THE EFFECT OF ALKALI ON THE ABSORPTION OF A PEPTIDE OF THYROXINE FROM THE GASTRO- INTESTINAL TRACT

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When pure thyroxine suspended in distilled water is administered by mouth it has only a slight effect on the basal metabolism (1). However, if pure thyroxine is dissolved in an excess of sodium hydroxide (presumably forming the disodium salt of thyroxine (2)) and is then administered by mouth, there is a well-marked increase in the basal metabolism which, on the average, is 63 per cent as great as the increase which follows the intravenous administration of thyroxine in the same form (3, 4). If the monosodium salt of thyroxine (tablets) is administered by mouth, the increase in basal metabolism is only 22 per cent as great as that which follows the intravenous injection of an equivalent dose of thyroxine in alkaline solution (1, 5). Thus an alkaline solution of thyroxine is nearly three times as effective, on the average, as an equivalent dose of the monosodium salt when both compounds are administered by mouth. Furthermore, it has been noted that the increase in basal metabolism produced by the oral administration of single large doses of thyroxine in alkaline solution is, on the average, about the same as that produced by single large doses of desiccated thyroid containing the same amount of iodine (6, 7).

By proteolytic digestion of thyroid, Harington and Salter (8) obtained a peptide of thyroxine which they state has "a much wider range of solubility" than thyroxine and contains thyroxine in a levorotatory form. Salter, Lerman and Means (9) have recently reported that this peptide had about the same effect whether given orally or intravenously and that it had the same effect as racemic thyroxine when both were administered intravenously in doses which contained the same total amounts of iodine. They have kindly informed us that, for oral administration, their peptide "was dissolved in sodium hydroxide" and the solution "then neutralized with phosphoric acid."

It seemed desirable to compare the effect of oral administration of this peptide with that of an equivalent amount of thyroxine in alkaline solution.

Accordingly, one of us (S. B. N.) prepared a similar substance by the method of Harington and Salter (8), using double the concentration of enzymes employed by them, i.e. 1.0 per cent pepsin and 0.4 per cent trypsin. From 1620 grams of desiccated thyroid¹ there were obtained 310 mgm. of a light buff colored powder containing 48 per cent iodine by the method of Leland and Foster (10) and 2.5 per cent total nitrogen by the micro-Kjeldahl method, giving a nitrogen:iodine ratio of 0.48:1. Before hydrolysis, 43 per cent of the total nitrogen was found to be in the amino form by the method of Folin (11); and, after six hours hydrolysis by the method of Harington and Salter (8), 78 per cent was found to be in this form. Most of the digestion products obtained by Harington and Salter contained from 45 to 50 per cent iodine and 5 per cent nitrogen, giving a nitrogen:iodine ratio of 1:1, but two of their products gave nitrogen:iodine ratios of 0.6:1. The product used by Salter, Lerman and Means (9) contained 49 per cent iodine and 3.3 per cent nitrogen, giving a nitrogen:iodine ratio of 0.6:1. When subjected to the action of nitrous acid and made alkaline with ammonia, the product we obtained gave an orange pink color in contrast to the red pink color which is characteristic of thyroxine. When it was dissolved in N/10 sodium hydroxide, the solution had a slight yellow tinge. In order to eliminate the possibility that it contained acid-soluble iodine, a small amount was dissolved in alkaline solution and precipitated at pH 5.0 by the cautious addition of dilute hydrochloric acid. Since 95 per cent of the iodine was recovered in the precipitate, it seemed logical to conclude that diiodotyrosine and inorganic iodine were both absent. From these data it would appear that our product is similar to the thyroxine peptide described by Harington and Salter.

Our peptide, like pure thyroxine, was insoluble in distilled water but was soluble in alkali. In view of this similarity in solubility it seemed all the more desirable to compare the calorogenic effects of oral administration of the two substances, (*a*) when suspended in distilled water, and (*b*) when dissolved in alkaline solution.

METHOD

The observations were made on three patients with well-marked myxedema. In the second patient the myxedema was spontaneous and in the other two it followed a subtotal thyroidectomy for exophthalmic goiter. Parts of the data (exclusive of those on the peptide) have been published elsewhere, as collected, to illustrate other points (1, 3, 7). The Sanborn-Benedict machine was used in the determinations of basal metabolism and

¹ The desiccated thyroid, pepsin and trypsin used in the preparation of this substance were very kindly supplied by Dr. Klein of the Wilson Laboratories, Chicago.

Aub-DuBois standards in the calculations. The number of calories produced by each type of treatment ("excess calories") has been calculated by a method previously described (3, 12). The synthetic thyroxine used was the crystalline powder purchased from Hoffmann-La Roche. The monosodium salt of synthetic thyroxine was bought from the same manufacturers in the form of tablets, each of which contained 1.03 mgm. of the salt.

DATA

The data are recorded in Charts 1 to 3 and summarized in Tables I and II. To facilitate comparisons in the tables, the effects of thyroxine have been calculated in terms of 10 mgm. (6.5 mgm. of iodine) and the effects of the peptide in terms of 13.5 mgm. (6.5 mgm. of iodine). It may be seen from Table I that, regardless of whether the effects of the various types of treatment are compared on the basis of the amount of increase in the basal metabolism or on the basis of the number of calories produced, similar conclusions are arrived at for the three patients in this study. Therefore, for the sake of simplicity, in discussing the data we shall confine our attention almost entirely to the amount of increase in the basal metabolism.

It may be noted that when the peptide was given by mouth suspended in distilled water, it had only about one-third as much effect as when it was given in alkaline solution. Thus, by calculation, the average increase in basal metabolism for a dose of 13.5 mgm. containing 6.5 mgm. of iodine was from minus 31 per cent to minus 22 per cent when the peptide was given suspended in distilled water and from minus 32 per cent to minus 7 per cent when it was given in alkaline solution. These increases in metabolism are nearly the same as those produced respectively by the oral administration of the monosodium salt of thyroxine in tablet form (from minus 33 per cent to minus 23 per cent, on the average) and by the oral administration of thyroxine in alkaline solution (from minus 32 per cent to minus 10 per cent, on the average) in doses which contained the same amounts of iodine. No adequate explanation can be offered for the similarity in the effect of oral administration of the peptide suspended in distilled water and that of the monosodium salt of thyroxine. It would appear that the peptide, when given suspended in distilled water, is absorbed as well as the monosodium salt in tablet form. If absorption depends upon the formation of a soluble salt in the small intestine, then it must follow that the peptide forms a soluble salt in this portion of the gastro-intestinal tract with greater ease than pure thyroxine, because, in the same dose by mouth, thyroxine as the free amino-acid does not produce a definite effect on the basal metabolism (1). It is possible that the effect of administering the peptide by mouth in an alkaline solution is slightly greater than that of administering thyroxine by mouth in an alkaline solution, but the data are not extensive enough to settle this point.

TABLE I
Comparison of calorigenic effects of thyroxine peptide with those of thyroxine and desiccated thyroid

Patient	Medication	Total iodine content of substance used	Basal metabolic rate before medication	Level to which basal metabolic rate rose	Change in basal metabolic rate	Time required for maximum change in basal metabolic rate	Length of time basal metabolic rate was affected	Time occupied by descending portion of metabolic curve	Number of squares (from charts for calculating "excess calories")	Number of "excess calories" produced	Change in terms of response to intravenous injection of 10 mgm. of thyroxine in alkaline solution	
											On basis of increase in basal metabolic rate	On basis of "excess calories" produced
		mgm.	per cent normal	per cent normal	points	days	days	days			per cent	per cent
Mrs. B. G. Lab. No. 2767	10 mgm. synthetic thyroxine in form of its monosodium salt (tablets) by mouth	6.5	-37	-22	15	8	50	28	426	6,485		
	10 mgm. synthetic thyroxine in form of its monosodium salt (tablets) in alkaline solution by mouth	6.5	-34	-17	17	6	52	32	580	8,910		
	50 mgm. pure synthetic thyroxine suspended in distilled water by duodenum	32.5	-32	-29	3	6	15	4	31	485		
	Calculated effect of 10 mgm. pure thyroxine suspended in distilled water by duodenum	6.5	-32	-31	1					97		
	10 mgm. synthetic thyroxine in alkaline solution by mouth	6.5	-34	-10	24	8	62	48	837	13,025		
	14.4 mgm. thyroxine peptide suspended in distilled water by mouth	6.9	-33	-22	11	3	41	30	282	4,410		
	Calculated effect of 13.5 mgm. thyroxine peptide suspended in distilled water by mouth	6.5	-33	-23	10					4,135		
	13.5 mgm. thyroxine peptide in alkaline solution by mouth	6.5	-36	-8	28	8	54	43	677	10,535		

TABLE I (continued)

Patient	Medication	Total iodine content of substance used	Basal metabolic rate before medication	Level to which basal metabolic rate rose	Change in basal metabolic rate	Time required for time basal metabolic rate was affected	Length of time basal metabolic rate was affected	Time occupied by dependent portion of metabolism curve	Number of squares (from chart for calculating "excess calories")	Number of "excess calories" produced	Change in terms of response to intravenous injection of 10 mgm. of thyroxine in alkaline solution	
											On basis of increase in basal metabolic rate	On basis of "excess calories" produced
		mgm.	per cent normal	per cent normal	points	days	days	days			per cent	per cent
Mrs. M. K. Lab. No. 2040	10 mgm. synthetic thyroxine in alkaline solution intravenously	6.5	-26	-1	25	2	29	22	408	5,135		
	10 mgm. Squibb's thyroxine in alkaline solution intravenously	6.5	-23	-4	19	4	29	17	342	4,300		
	Average effect of 10 mgm. thyroxine in alkaline solution intravenously	6.5	-25	-3	22	3	29	20		4,720	100	100
	10 mgm. synthetic thyroxine in form of the monosodium salt (tablets) by mouth	6.5	-25	-19	6	3	25	14	102	1,305	27	28
	10 mgm. synthetic thyroxine in alkaline solution by mouth	6.5	-26	-6	20	6	33	24	332	4,315	91	91
	2.75 grams (42.4 grains) desiccated thyroid (tablets) by mouth	6.3	-26	-11	15	2	33	26	248	3,170		
	Calculated effect of 2.83 grams desiccated thyroid by mouth	6.5	-26	-11	15					3,260	68	69
	14.4 mgm. thyroxine peptide suspended in distilled water by mouth	6.9	-26	-17	9	2	20	17	114	1,455		
	Calculated effect of 13.5 mgm. thyroxine peptide suspended in distilled water by mouth	6.5	-26	-18	8					1,365	36	29
	13.5 mgm. thyroxine peptide in alkaline solution by mouth	6.5	-28	-6	22	2	31	21	365	4,690	100	99

TABLE I (continued)

Patient	Medication	Total iodine content of substance used	Basal metabolic rate before medication	Level to which basal metabolic rate rose	Change in basal metabolic rate	Time required for maximum change in basal metabolic rate	Length of time basal metabolic rate was affected	Time occupied by dependent portion of metabolism curve	Number of squares (from charts for calculating "excess calories")	Number of "excess calories" produced	Change in terms of response to intravenous injection of 10 mgm. of thyroxine in alkaline solution	
											On basis of increase in basal metabolic rate	On basis of "excess calories" produced
		mgm.	per cent normal	per cent normal	points	days	days	days			per cent	per cent
Mrs. A. R. MacN. No. 1000	7.5 mgm. synthetic thyroxine in alkaline solution intravenously	4.9	-30	-7	23	6	45	32	476	6,910		
	7.5 mgm. Squibb's thyroxine in alkaline solution intravenously	4.9	-34	-9	25	7	53	41	623	9,000		
	Average effect of 7.5 mgm. thyroxine in alkaline solution intravenously	4.9	-32	-8	24	6	49	37		7,955		
	Calculated effect of 10 mgm. thyroxine in alkaline solution intravenously	6.5	-32	0	32					10,605	100	100
	7.5 mgm. synthetic thyroxine in form of its monosodium salt (tablets) by mouth	4.9	-36	-29	7	6	27	10	125	1,825		
	Calculated effect of 10 mgm. synthetic thyroxine in form of its monosodium salt by mouth	6.5	-36	-27	9					2,435	28	23
	7.5 mgm. synthetic thyroxine in alkaline solution by mouth	4.9	-35	-19	16	5	42	28	342	4,980		
	Calculated effect of 10 mgm. synthetic thyroxine in alkaline solution by mouth	6.5	-35	-14	21					6,655	66	63
	2.05 grams (31.7 grains) desiccated thyroid (tablets) by mouth	4.7	-35	-21	14	2	34	31	299	4,385		
	Calculated effect of 2.83 grams desiccated thyroid by mouth	6.5	-35	-16	19					6,055	59	57
	10.9 mgm. thyroxine peptide suspended in distilled water by mouth	5.2	-34	-26	8	1	28	26	149	2,205		
	Calculated effect of 13.5 mgm. thyroxine peptide suspended in distilled water by mouth	6.5	-34	-24	10					2,730	31	26
	10.2 mgm. thyroxine peptide in alkaline solution by mouth	4.9	-34	-14	20	3	47	42	506	7,455		
	Calculated effect of 13.5 mgm. thyroxine peptide in alkaline solution by mouth	6.5	-34	-8	26					9,865	81	93

TABLE I (continued)

Patient	Medication	Total iodine content of substance used	Basal metabolic rate before medication	Level to which basal metabolic rate rose	Change in basal metabolic rate	Time required for maximum change in basal metabolic rate	Length of time basal metabolic rate was affected	Time occupied by descending portion of metabolism curve	Number of squares (from chart for calculating "excess calories")	Change in terms of response to intravenous injection of 10 mgm. of thyroxine in alkaline solution	
										On basis of increase in basal metabolic rate	On basis of "excess calories" produced
		mgm.	per cent normal	per cent normal	points	days	days	days		per cent	per cent
	<i>Aeroges</i>										
	10 mgm. synthetic thyroxine in form of its monosodium salt (tablets) by mouth										
	All three patients.....	6.5	-33	-23	10					30	24
	Last two patients.....	6.5	-31	-23	8						
	10 mgm. synthetic thyroxine in alkaline solution by mouth										
	All three patients.....	6.5	-32	-10	22					78	72
	Last two patients.....	6.5	-31	-10	21						
	13.5 mgm. thyroxine peptide suspended in distilled water by mouth										
	All three patients.....	6.5	-31	-22	9					33	27
	Last two patients.....	6.5	-30	-21	9						
	13.5 mgm. thyroxine peptide in alkaline solution by mouth										
	All three patients.....	6.5	-32	-7	25					89	95
	Last two patients.....	6.5	-31	-7	24						
	2.83 grms. decaecated thyroid (tablets) by mouth										
	Last two patients.....	6.5	-31	-14	17					63	61
	10 mgm. thyroxine in alkaline solution Intravenously										
	Last two patients.....	6.5	-29	-2	27					100	100

TABLE II
Summary of effects of various compounds of thyroxine

Medication	Total iodine content of substance used	Number of patients	Number of administrations	Average basal metabolic rate before treatment	Average level to which basal metabolic rate rose	Average change in basal metabolic rate	Average number of "excess calories" produced	Change in terms of average response to intravenous injection of 10 mgm. thyroxine in alkaline solution	
								On basis of increase in basal metabolic rate	On basis of increase in "excess calories" produced
	mgm.			per cent normal	per cent normal	points		per cent	per cent
10 mgm. ¹ pure synthetic thyroxine suspended in distilled water, by duodenum.....	6.5	3	3	-30	-29	1		3	
10 mgm. ² pure synthetic thyroxine suspended in distilled water, by mouth.....	6.5	4	4	-32	-30	2		6	
10 mgm. ³ synthetic thyroxine in form of monosodium salt (tablets), by mouth.....	6.5	6	6	-28	-21	7		22	
Patients receiving monosodium salt by mouth, in whom "excess calories" were calculated.....	6.5	4	4	-30	-22	8	2,755	25	18
13.5 mgm. thyroxine peptide suspended in distilled water, by mouth.....	6.5	3	3	-31	-22	9	2,745	28	18
10 mgm. ⁴ synthetic thyroxine in alkaline solution, by mouth.....	6.5	5	5	-31	-11	20		63	
Patients receiving thyroxine in alkaline solution by mouth, in whom "excess calories" were calculated.....	6.5	4	4	-34	-12	22	9,010	69	58
2.83 grams, ⁵ desiccated thyroid (tablets) by mouth.....	6.5	5	5	-37	-15	22	7,405	69	48
13.5 mgm. thyroxine peptide in alkaline solution, by mouth.....	6.5	3	3	-32	-7	25	8,365	78	54
10 mgm. ⁶ thyroxine in alkaline solution (sodium or potassium hydroxide), intravenously.....	6.5	6	8	-37	-5	32	15,520	100	100

¹ The doses used were 10 mgm., 40 mgm. and 50 mgm.

² The doses used were 10 mgm., 30 mgm., 40 mgm. and 100 mgm.

³ The doses used were four of 10 mgm. each, one of 30 mgm. and one of 40 mgm.

⁴ All doses were 10 mgm. except one of 7.5 mgm.

⁵ All doses were 2.75 grams except one of 2.05 grams.

⁶ All doses were 10 mgm. except two of 7.5 mgm. Sodium hydroxide was used to dissolve the thyroxine for all administrations except one, in which potassium hydroxide was used.

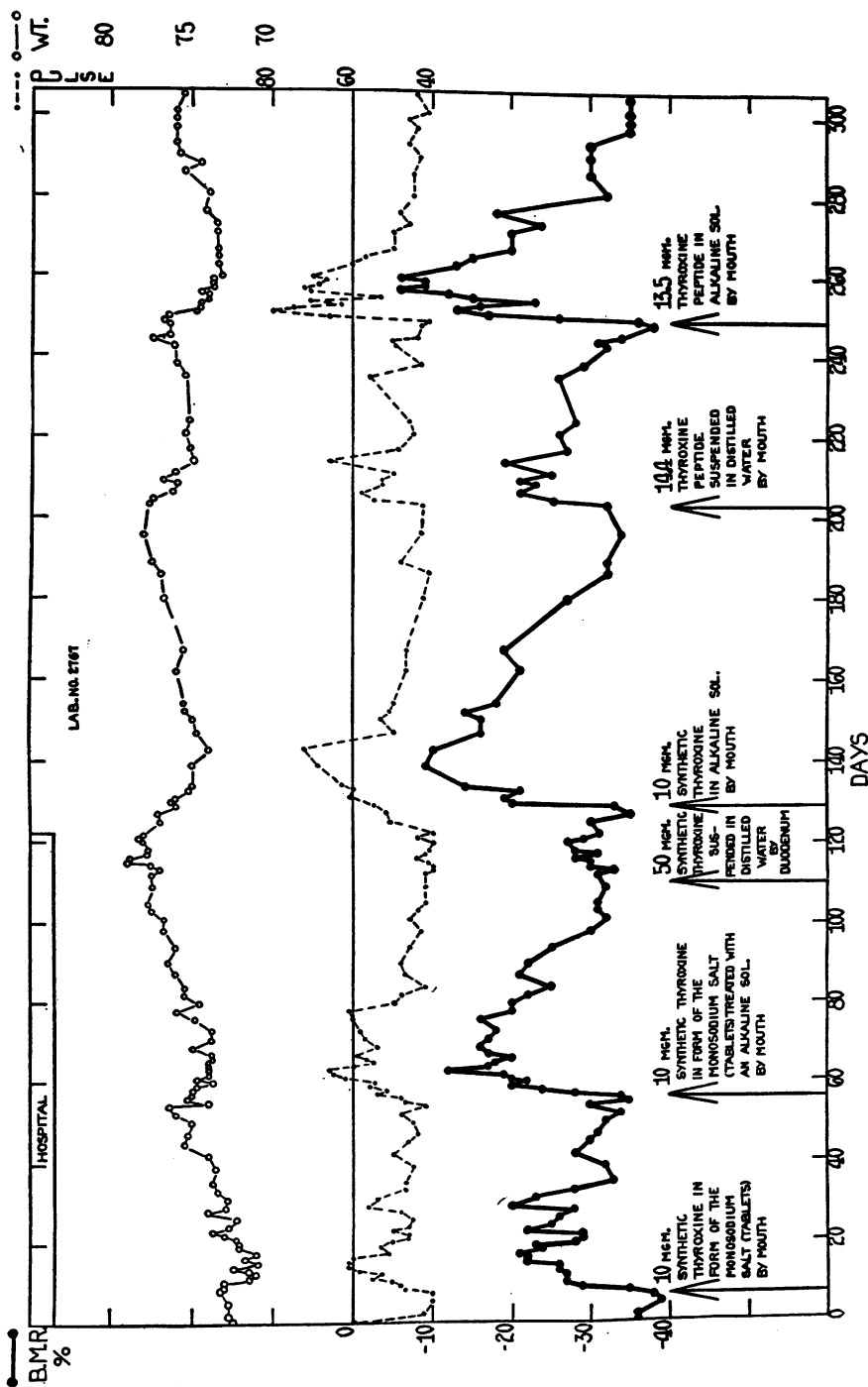


CHART 1. MRS. B. G. HEIGHT 158 CM. AGE 38

Comparison of the effects of oral administration of thyroxine peptide suspended in distilled water and in alkaline solution with those of administering thyroxine in various forms by mouth, and intravenously in alkaline solution.

For details of the administration of pure thyroxine and of its sodium salts, see another publication (1). The single dose of 14.4 mgm. of thyroxine peptide was suspended in distilled water and administered by mouth at 2.35 p.m., May 26, 1933, a total of 500 cc. of distilled water being used, largely for rinsing. The patient had had no breakfast or lunch. The single dose of 13.5 mgm. of thyroxine peptide in alkaline solution was administered by mouth at 2.25 p.m., July 11, 1933, a total of 10 drops of 10 per cent sodium hydroxide and 500 cc. of distilled water being used for solution and rinsing. The patient had had no breakfast or lunch.

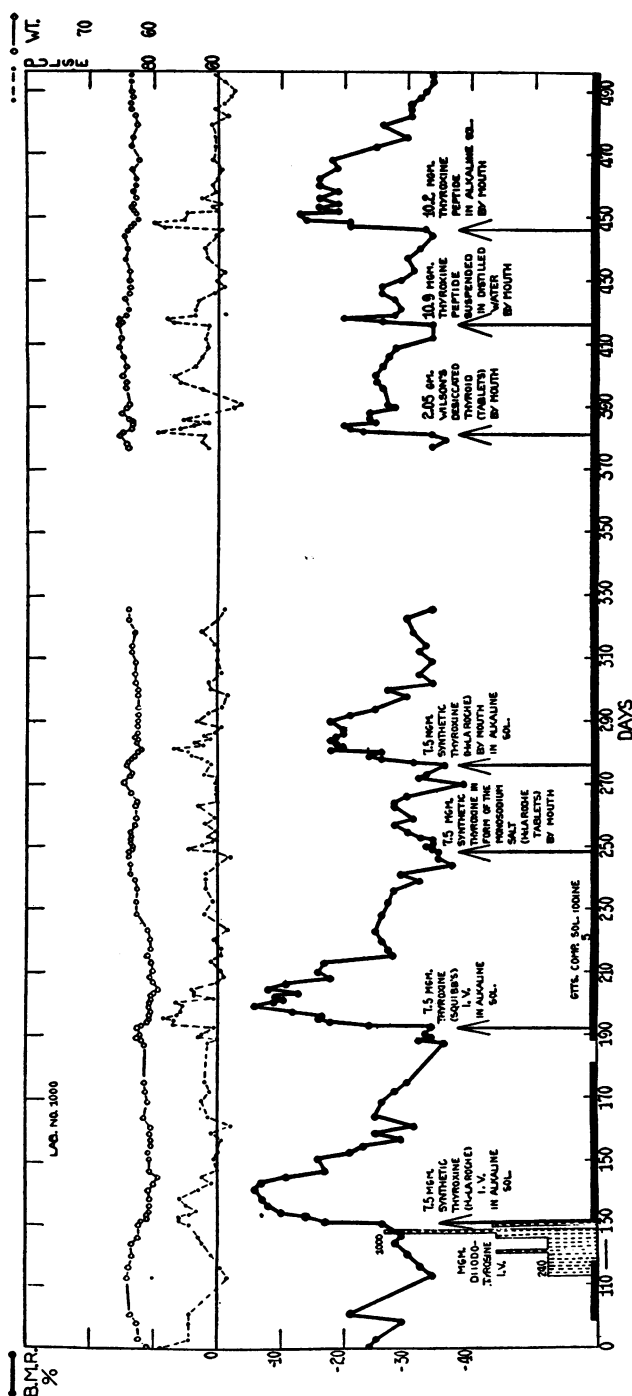


CHART 3. MRS. A. R. HEIGHT 160 CM. AGE 33

Comparison of the effects of oral administration of thyroxine peptide suspended in distilled water and in alkaline solution with those of administering thyroxine in various forms by mouth, and intravenously in alkaline solution.

For details of the various administrations of monosodium thyroxine, thyroxine in alkaline solution, and desiccated thyroid, see other publications (3, 7). The single dose of 10.9 mgm. of thyroxine peptide was administered by mouth, suspended in distilled water, at 12 o'clock (noon), May 29, 1933, a total of 500 cc. of water being used, largely for rinsing. The single dose of 10.2 mgm. of thyroxine peptide in alkaline solution was administered by mouth at 3.35 p.m., June 28, 1933, a total of 9 drops of 10 per cent sodium hydroxide and 500 cc. of distilled water being used for solution and rinsing. Preceding both administrations of the peptide, the patient had not eaten breakfast or lunch.

In the second and third patients we have also observed the effects of administering single large doses of thyroxine in alkaline solution intravenously and equivalent doses of desiccated thyroid by mouth. In these two patients it may be calculated that for doses containing 6.5 mgm. of iodine, the intravenous administration of thyroxine in alkaline solution produced an average increase in the basal metabolism from minus 29 per cent to minus 2 per cent, and the oral administration of desiccated thyroid an increase from minus 31 per cent to minus 14 per cent, as compared with an increase from minus 31 per cent to minus 7 per cent for oral administration of the peptide in alkaline solution. The average increases produced by oral administration of the peptide suspended in distilled water and the monosodium salt of thyroxine were nearly the same for these two patients as for all three patients. Thus, the effect of administering single

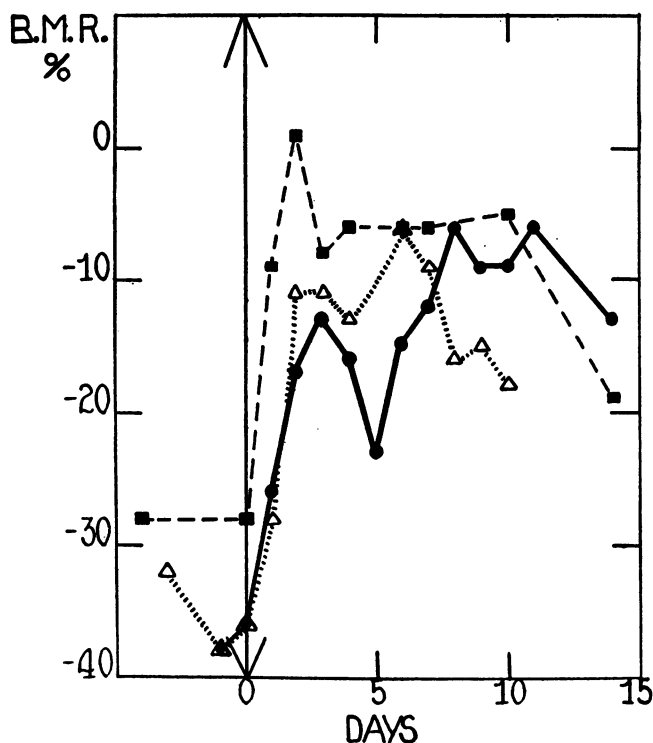


CHART 4

Comparison of the effect on the basal metabolism observed by Salter, Lerman and Means (9) from oral administration of 13.0 mgm. of their peptide containing 6.5 mgm. of iodine to one patient with myxedema (white triangles) with that observed by us from oral administration of an alkaline solution of 13.5 mgm. of our peptide containing 6.5 mgm. of iodine to two patients with myxedema (black circles, the first patient; black squares, the second patient).

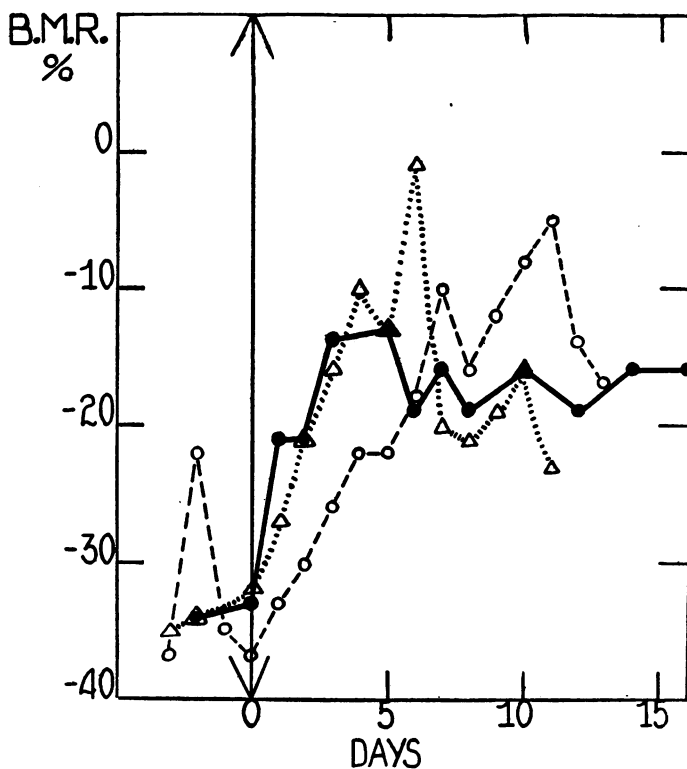


CHART 5

Comparison of the effects on basal metabolism observed by Salter, Lerman and Means (9) from oral administration of 10.0 mgm. of their peptide containing 4.9 mgm. of iodine to two patients with myxedema (interrupted lines) with those observed by us from oral administration of an alkaline solution of 10.2 mgm. of our peptide containing 4.9 mgm. of iodine to one patient with myxedema (solid line, the third patient).

large doses of peptide by mouth in alkaline solution was nearly the same as that of administering single large doses of thyroxine in alkaline solution intravenously, but was greater than that of administering single large doses of desiccated thyroid by mouth. However, the effect of desiccated thyroid was about twice as great as that produced by oral administration of the peptide suspended in distilled water or that produced by oral administration of the monosodium salt of thyroxine.

Salter, Lerman and Means (9) have reported the effects of three separate oral administrations of their polypeptide—one of 13.0 mgm. containing 6.5 mgm. of iodine and two of 10.0 mgm. each containing 4.9 mgm. of iodine. By changing the effects to terms of 6.5 mgm. of iodine we have calculated that this dose of iodine administered by mouth in the form of

their polypeptide would have produced an increase in the basal metabolism from minus 36 per cent to minus 6 per cent on the average, a change which is nearly the same as that produced by the oral administration of 6.5 mgm. of iodine in the form of our peptide (13.5 mgm.) in alkaline solution. We have compared our data with theirs in Charts 4 and 5. It is of interest that their peptide was dissolved in an alkaline solution and the solution then neutralized.

It is of interest to combine the data of the present study with those which we have previously reported concerning the effects of the administration of thyroxine in various forms by the oral and intravenous routes. This has been done in Table II. It may be noted that, on the average, the increases in basal metabolism produced by the oral administration of single large doses of monosodium thyroxine, thyroxine peptide suspended in distilled water, pure thyroxine in alkaline solution, desiccated thyroid and thyroxine peptide in alkaline solution are respectively 22 per cent, 28 per cent, 63 per cent, 69 per cent and 78 per cent as great as those produced by the intravenous injection of an alkaline solution of single large doses of pure thyroxine containing the same amount of iodine. In terms of "excess calorie" production, the corresponding figures are 18, 18, 58, 48 and 54 per cent respectively.

SUMMARY

From a proteolytic digest of desiccated thyroid we have prepared a peptide of thyroxine containing 48 per cent iodine, with a nitrogen:iodine ratio of 0.48:1. This product is insoluble in distilled water but soluble in a dilute solution of sodium hydroxide.

When suspended in distilled water and administered by mouth to patients with myxedema it produced only a slight increase in the basal metabolism, which was about the same as that produced by oral administration of the monosodium salt of thyroxine in doses which contained the same amounts of iodine, and about one-quarter as great as that produced by thyroxine in alkaline solution given intravenously. However, when administered by mouth in an alkaline solution, the peptide produced a well-marked increase in basal metabolism which was nearly four-fifths as great as that produced by thyroxine in alkaline solution given intravenously, and slightly greater than those produced by oral administration of desiccated thyroid and thyroxine in alkaline solution.

BIBLIOGRAPHY

1. Thompson, W. O., Thompson, P. K., Taylor, S. G., III, and Dickie, L. F. N., Oral and duodenal administration of single large doses of pure thyroxine: Comparison of calorogenic effects with those of monosodium thyroxine and thyroxine in alkaline solution. *Arch. Int. Med.* (In press.)

2. Kendall, E. C., Thyroxine. The Chemical Catalog Company, Inc., New York, 1929.
3. Thompson, W. O., Thompson, P. K., Dickie, L. F. N., and Alper, J. M., Effect of alkali on the absorption of thyroxine from the gastro-intestinal tract, with a note on the comparative effects of synthetic and "natural" thyroxine injected intravenously. *Arch. Int. Med.*, 1933, **52**, 809.
4. Thompson, W. O., Thompson, P. K., Taylor, S. G., III, Alper, J. M., and Dickie, L. F. N., The effects of various compounds of thyroxine on the basal metabolism. *Endocrinology*, 1934, **18**, 228.
5. Thompson, W. O., Thompson, P. K., and Dickie, L. F. N., Monosodium thyroxine, desiccated thyroid and an impure sodium salt of thyroxine. Comparison of their effects when administered orally with the effect of thyroxine injected intravenously in an alkaline solution. *Arch. Int. Med.*, 1933, **52**, 576.
6. Thompson, W. O., Thompson, P. K., Dickie, L. F. N., and Taylor, S. G., III, The iodine in the thyroid gland. *West. J. Surg.*, 1933, **41**, 431.
7. Thompson, W. O., Thompson, P. K., Taylor, S. G., III, and Dickie, L. F. N., The calorigenic action of single large doses of desiccated hog thyroid: Comparison with that of thyroxine given orally and intravenously. *Arch. Int. Med.* (In press.)
8. Harington, C. R., and Salter, W. T., The isolation of l-thyroxine from the thyroid gland by the action of proteolytic enzymes. *Biochem. J.*, 1930, **24**, 456.
9. Salter, W. T., Lerman, J., and Means, J. H., The calorigenic action of thyroxin polypeptide. *J. Clin. Invest.*, 1933, **12**, 327.
10. Leland, J. P., and Foster, G. L., A method for the determination of thyroxine in the thyroid. *J. Biol. Chem.*, 1932, **95**, 165.
11. Folin, O., with the assistance of Wu, H., A system of blood analysis. Supplement III. A new colorimetric method for the determination of the amino-acid nitrogen in blood. *J. Biol. Chem.*, 1922, **51**, 377.
12. Thompson, W. O., Thompson, P. K., Brailey, A. G., and Cohen, A. C., The calorigenetic action of thyroxin at different levels of basal metabolism in myxedema. *J. Clin. Invest.*, 1929, **7**, 437.